ALTERNATIVE SOURCE DEMONSTRATION REPORT FEDERAL CCR RULE

H.W. Pirkey Power Plant East Bottom Ash Pond Hallsville, Texas

Submitted to



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LIST OF ACRONYMS

AEP American Electric Power

ASD Alternative Source Demonstration

CCR Coal Combustion Residuals

CFR Code of Federal Regulations

EBAP East Bottom Ash Pond

EDS Energy Dispersive Spectroscopic Analyzer

EPRI Electric Power Research Institute

GSC Groundwater Stats Consulting, LLC

GWPS Groundwater Protection Standard

LCL Lower Confidence Limit

MCL Maximum Contaminant Level

QA Quality Assurance

QC Quality Control

SEM Scanning Electron Microscopy

SPLP Synthetic Precipitation Leaching Profile

SSL Statistically Significant Level

TCEQ Texas Commission on Environmental Quality

UTL Upper Tolerance Limit

USEPA United States Environmental Protection Agency

VAP Vertical Aquifer Profiling

WBAP West Bottom Ash Pond

XRD X-Ray Diffraction

SECTION 1

INTRODUCTION AND SUMMARY

This Alternative Source Demonstration (ASD) report has been prepared to address statistically significant levels (SSLs) for cobalt and lithium in the groundwater monitoring network at the H.W. Pirkey Plant East Bottom Ash Pond (EBAP), located in Hallsville, Texas, following the second semi-annual assessment monitoring event of 2020. The EBAP is registered as a surface impoundment under Texas Commission on Environmental Quality (TCEQ) Industrial and Hazardous Waste Solid Waste Registration No. 33240.

The H.W. Pirkey Plant has four regulated coal combustion residuals (CCR) storage units, including the EBAP (**Figure 1**). In November 2020, a semi-annual assessment monitoring event was conducted at the EBAP in accordance with 40 CFR 257.95(d)(1). The monitoring data were submitted to Groundwater Stats Consulting, LLC (GSC) for statistical analysis. Groundwater protection standards (GWPSs) were established for each Appendix IV parameter in accordance with the statistical analysis plan developed for the unit (Geosyntec, 2020a) and the United States Environmental Protection Agency's (USEPA's) *Statistical Analysis of Groundwater Monitoring Data at RCRA Facilities – Unified Guidance* (Unified Guidance; USEPA, 2009). The GWPS for each parameter was established as the greater of the background concentration and the maximum contaminant level (MCL) or, for parameters without an MCL, the risk-based level specified in 40 CFR 257.95(h)(2). To determine background concentrations, an upper tolerance limit (UTL) was calculated using pooled data from the background wells collected during the background monitoring and assessment monitoring events.

Confidence intervals were re-calculated for each Appendix IV parameter at the compliance wells to assess whether these parameters were present at a statistically significant level (SSL) above the GWPSs. An SSL was concluded if the lower confidence limit (LCL) of a parameter exceeded the GWPS (i.e., if the entire confidence interval exceeded the GWPS). The following SSLs were identified at the Pirkey EBAP (Geosyntec, 2021):

- The LCLs for cobalt exceeded the GWPS of 0.00940 mg/L at AD-2 (0.0100 mg/L), AD-31 (0.00953 mg/L), and AD-32 (0.0239 mg/L).
- The LCL for lithium exceeded the GWPS of 0.0590 mg/L at AD-31 (0.0835 mg/L) and AD-32 (0.0838 mg/L).

No other SSLs were identified.

1.1 CCR Rule Requirements

USEPA regulations regarding assessment monitoring programs for CCR landfills and surface impoundments provide owners and operators with the option to make an alternative source demonstration when an SSL is identified (40 CFR 257.95(g)(3)(ii)). An owner or operator may:

Demonstrate that a source other than the CCR unit caused the contamination, or that the statistically significant increase resulted from error in sampling, analysis, statistical evaluation, or natural variation in groundwater quality. Any such demonstration must be supported by a report that includes the factual or evidentiary basis for any conclusions and must be certified to be accurate by a qualified professional engineer or approval from the Participating State Director or approval from EPA where EPA is the permitting authority. If a successful demonstration is made, the owner or operator must continue monitoring in accordance with the assessment monitoring program pursuant to this section...

Pursuant to 40 CFR 257.95(g)(3)(ii), Geosyntec Consultants, Inc. (Geosyntec) has prepared this ASD report to document that the SSLs identified for cobalt and lithium are from a source other than the EBAP.

1.2 <u>Demonstration of Alternative Sources</u>

An evaluation was completed to assess possible alternative sources to which the identified SSLs could be attributed. Alternative sources were identified amongst five types, based on methodology provided by EPRI (2017):

- ASD Type I: Sampling Causes;
- ASD Type II: Laboratory Causes;
- ASD Type III: Statistical Evaluation Causes;
- ASD Type IV: Natural Variation; and
- ASD Type V: Alternative Sources.

A demonstration was conducted to show that the SSLs identified for cobalt and lithium were based on a Type IV cause and not by a release from the Pirkey EBAP.

SECTION 2

ALTERNATIVE SOURCE DEMONSTRATION

The Federal CCR Rule allows the owner or operator 90 days from the determination of an SSL to demonstrate that a source other than the CCR unit caused the SSL. The methodology used to evaluate the SSLs identified for cobalt and lithium and the proposed alternative sources are described below.

2.1 Proposed Alternative Source

An initial review of site geochemistry, site historical data, and laboratory quality assurance/quality control (QA/QC) data did not identify alternative sources for cobalt and lithium due to Type I (sampling), Type II (laboratory), or Type III (statistical evaluation) issues. Groundwater sampling, laboratory analysis, and statistical evaluations were generally completed in accordance with the Federal CCR Rule and draft TCEQ guidance for groundwater monitoring (TCEQ, 2020). As described below, the SSLs have been attributed to natural variation associated with the underlying geology, which is a Type IV (natural variation) issue.

2.1.1 Cobalt

Previous ASDs for cobalt at the EBAP provided evidence that cobalt is present in the aquifer media at the site and that the observed cobalt concentrations were due to natural variation (Geosyntec, 2019a; Geosyntec, 2019b; Geosyntec, 2020b; Geosyntec, 2020c). The previous ASDs discussed how the EBAP did not appear to be a source for cobalt in downgradient groundwater, based on observed concentrations of cobalt both in the ash material and in leachate from Synthetic Precipitation Leaching Procedure (SPLP) analysis (SW-846 Test Method 1312, [USEPA, 1994]) of the ash material. Cobalt was not detected in the SPLP ash leachate above the reporting limit of 0.01 mg/L, which is lower than the average concentrations observed at the wells of interest (**Table 1**).

A surface water sample was collected directly from the WBAP on November 4, 2020 as a surrogate for an EBAP sample. A sample could not be collected from the EBAP, as all ponded water had been removed at the time of sampling. However, the EBAP and WBAP receive the same process water, with the use of each pond dependent on available freeboard and cleaning schedule; thus, there is a basis for the equivalency between these two surface water samples. Cobalt was detected at a concentration of 0.000501 mg/L in the WBAP sample (**Table 1**). Cobalt was detected in a surface water sample previously collected (December 15, 2018) from the EBAP at an estimated concentration of 0.0024 mg/L (**Table 1**). These concentrations are lower than all reported cobalt concentrations for in network wells from the most recent sampling event (excluding upgradient well AD-18), and over an order of magnitude lower than the average concentration in groundwater at the wells of interest (**Table 1**; **Figure 2**). Thus, the EBAP is not the likely source of cobalt at AD-2, AD-31, and AD-32.

As noted in the previous ASDs, soil samples collected across the site, including from locations near the EBAP, identified cobalt in the aquifer solids at varying concentrations. SB-2 was advanced in the vicinity of AD-2 in April 2020 to re-log the geology at AD-2 and collect samples for laboratory analysis of total metals and mineralogy. The SB-2 field boring log, which was generated by Auckland Consulting LLC, is provided as **Attachment A**. Cobalt was identified at SB-2 at concentrations of 9.45 milligrams per kilogram (mg/kg) at 25-27 feet below ground surface (bgs) and 19.2 mg/kg at 31-33 feet bgs (**Table 2**). These cobalt concentrations are greater than the concentration of cobalt present in the bottom ash (**Table 1**). Both samples correlate to the depth of the monitoring well screen of AD-2 (20-40 feet bgs), indicating that cobalt is present in aquifer solids within the AD-2 screened interval. Cobalt was also identified in the aquifer solids at varying concentrations at other locations throughout the site, with the highest value of 23.5 mg/kg reported at AD-41, which is upgradient of the EBAP (**Figure 3**).

In addition to total cobalt, soil samples were submitted for mineralogical analysis to evaluate the presence of cobalt-containing minerals. X-ray diffraction (XRD) analysis of soils from SB-2 identified pyrite (an iron sulfide) in samples collected at 25-27 feet bgs and 31-33 feet bgs at concentrations up to 7% by weight (**Figure 3**). Cobalt is known to undergo isomorphic substitution for iron in crystalline iron minerals such as pyrite due to their similar ionic radii of approximately 1.56 angstroms (Å) for iron vs. 1.52 Å for cobalt (Clementi and Raimondi, 1963; Krupka and Serne, 2002; Hitzman et al., 2017).

The aquifer solids at SB-2 are distinctly red in color at shallow depths, as illustrated in the photolog of soil cores provided in **Attachment B**. While shallow samples were not collected for mineralogical analysis, red color in soils is often associated with the presence of oxidized iron-bearing minerals such as hematite and goethite. The weathering of pyrite to goethite under oxidizing conditions is also a well-understood phenomenon, including in formations in east Texas (Senkayi et al., 1986; Dixon et al., 1982). It is likely that the pyrite weathering process is resulting in the release of isomorphically substituted cobalt from the pyrite crystal structure as it undergoes oxidative transformation to iron oxide minerals.

As described in the previous ASDs, vertical aquifer profiling (VAP) was used to collect groundwater samples from upgradient locations B-2 and B-3 during the soil boring and sample collection process (Geosyntec, 2019b). A groundwater sample was also collected from AD-32, an existing well within the EBAP groundwater monitoring network. Solid phases within these groundwater samples were separated and submitted for analysis of chemical composition. For the VAP samples, separation was completed using a centrifuge due to the high abundance of solids. For the groundwater sample at AD-32, the sample was filtered using a 1.5-micron filter. Based on total metals analysis, cobalt was identified both in the centrifuged solid material collected from upgradient VAP location B-3 [VAP-B3-(40-45)] and in the material retained on the filter after processing groundwater from permanent monitoring wells B-2 and B-3 (**Table 2**). The concentrations of cobalt in the solid material retained after filtration were comparable to the bulk soil samples collected from the same locations.

The solid sample [VAP-B3-(40-45)] was submitted for mineralogical analysis via XRD and scanning electron microscopy (SEM) using an energy dispersive spectroscopic analyzer (EDS). The XRD results identified pyrite as approximately 3% of the solid phase (**Table 3**). Pyrite was identified during SEM/EDS analysis of lignite which is mined immediately adjacent to the site. Logging completed while the VAP boring was advanced identified coal at several intervals, including 45 and 48 feet bgs (**Figure 4**). Furthermore, SEM/EDS of both centrifuged solid samples [VAP-B3-(40-45) and VAP-B3-(50-55)] identified pyrite in backscattered electron micrographs by the distinctive framboidal morphology (Harris et al., 1981; Sawlowicz, 2000). Major peaks involving iron and sulfur were identified in the EDS spectrum, which further support the identification of pyrite (**Attachment C**). While cobalt was not identified in the EDS spectrum, it is likely present at concentrations below the detection limit.

Naturally occurring cobalt is known to substitute for iron in pyrite, which is then known to weather to iron oxides. The presence of pyrite has been confirmed at AD-2 and across the Site. This suggests that pyrite may be providing a source for aqueous cobalt in groundwater. Additionally, the pond was not identified as the source of cobalt at wells in the EBAP network based on the low concentrations of cobalt in the pond itself.

2.1.2 Lithium

Previous ASDs for lithium at the EBAP attributed the observed lithium exceedances to variations in naturally suspended aquifer solids that likely originate from naturally occurring lignite and are ubiquitous in the aquifer based on the presence of lithium at upgradient locations and in the solid phase (Geosyntec, 2019b; Geosyntec, 2020b; Geosyntec, 2020c). Data gathered in support of the prior ASDs and recent results provide additional evidence that the observed lithium concentrations at AD-31 and AD-32 are due to natural variation in the aquifer.

As discussed in Section 2.1.1, a surface water sample was collected directly from the WBAP on November 4, 2020, as a surrogate for an EBAP sample. Lithium was detected in the WBAP sample at a concentration of 0.0274 mg/L, which is comparable to the estimated concentration of 0.023 mg/L reported for the EBAP water in 2018 (**Table 4**). The mobile fraction identified in the bottom ash by SPLP was even lower, with an estimated lithium concentration of 0.011 mg/L. These concentrations are lower than the average lithium concentrations at AD-31 and AD-32 (**Table 4**). Thus, the EBAP is not the likely source of lithium at AD-31 and AD-32.

Groundwater samples collected from upgradient wells B-2 and B-3 in March 2021 had total lithium concentrations of 0.0610 mg/L and 0.0686 mg/L, respectively, both of which were above the GWPS of 0.0590 mg/L (**Figure 5**). Because B-2 and B-3 were installed at locations upgradient to and unimpacted by site activities, their lithium concentrations suggest that lithium is naturally present at concentrations above the GWPS in the vicinity of the EBAP. It is noted that B-2 and B-3 are not part of the monitoring network for the EBAP, and as such the lithium concentrations in groundwater from these wells are not considered in calculating the GWPS for the CCR unit.

As described in Section 2.1.1, groundwater samples were collected from B-2, B-3, and AD-32 and filtered to separate solids. Groundwater was also collected from a VAP boring (VAP-B3-(40-45)) and centrifuged to separate solids. Lithium was detected in the solid material separated from these groundwater samples at concentrations comparable to bulk soil at all locations, providing evidence that the particulates captured during groundwater sampling contain lithium (**Table 5**).

2.1.2.1 Calculated Partition Coefficients

A previous ASD for lithium at the EBAP developed a proposed lithium mobility in groundwater due to desorption from clay minerals associated with naturally occurring lignite material. This mechanism was posited as the source of lithium in both upgradient and downgradient wells at the EBAP (Geosyntec, 2019b). Previously completed XRD analysis of centrifuged solid material samples (VAP-B3-(40-45)) found that clay minerals, including kaolinite, smectite, and illite/mica, made up at least 60% of the aquifer solid (**Table 3**). These clay minerals, particularly smectite and illite, are known to retain positively charged ions such as lithium via cation exchange processes. SEM/EDS analysis identified the presence of silicon, aluminum and oxygen, all of which are indicative of clay minerals (**Attachment A**). The backscattered electron micrographs of these samples also identified clay particles by morphology. The largest clay particles (> 5 μm) are likely kaolinite, while smectite and illite dominate the smaller size fraction.

Total metal concentrations in the solid materials separated from the groundwater samples during filtration and the filtered groundwater concentrations were used to calculate partition coefficients values (K_d) for lithium, potassium, and sodium. Details about the K_d calculation are provided in the previous ASD (Geosyntec, 2019b). K_d values for groundwater and particulates collected from wells B-2, B-3, and AD-32 were comparable to literature K_d values reported for organic-rich media such as bogs and peat beds (Sheppard et al., 2009; Sheppard et al., 2011), providing further evidence that lithium mobility in site groundwater is similar to other sites with organic-rich soils (**Table 6**). Additionally, the calculated K_d values for Pirkey soils were consistent with the literature, with potassium having the highest K_d (greatest affinity for sorption) and sodium the lowest K_d (least affinity for sorption). Furthermore, the values are similar for groundwater from all three wells, suggesting a universal mechanism controlling lithium, sodium, and potassium mobility in groundwater. Since the site-specific Kd values were calculated, lithium concentrations at the wells of interest have remained consistent, suggesting that this cation exchange mechanism is still controlling lithium groundwater concentrations (**Figure 6**).

These multiple lines of evidence show that elevated lithium concentrations at AD-31 and AD-32 are not due to a release from the EBAP, and instead can be attributed to natural variation. This variation appears related to the distribution of clay fractions associated with lignite materials in the soil aquifer material.

2.2 <u>Sampling Requirements</u>

As the ASD presented above supports the position that the identified SSLs are not due to a release from the Pirkey EBAP, the unit will remain in the assessment monitoring program. Groundwater at the unit will continue to be sampled for Appendix IV parameters on a semiannual basis.

SECTION 3

CONCLUSIONS AND RECOMMENDATIONS

The preceding information serves as the ASD prepared in accordance with 40 CFR 257.95(g)(3)(ii) and supports the position that the SSLs for cobalt and lithium identified during assessment monitoring in November 2020 were not due to a release from the EBAP. The identified SSLs were instead attributed to natural variation in the underlying geology. Therefore, no further action is warranted, and the Pirkey EBAP will remain in the assessment monitoring program. Certification of this ASD by a qualified professional engineer is provided in **Attachment E.**

SECTION 4

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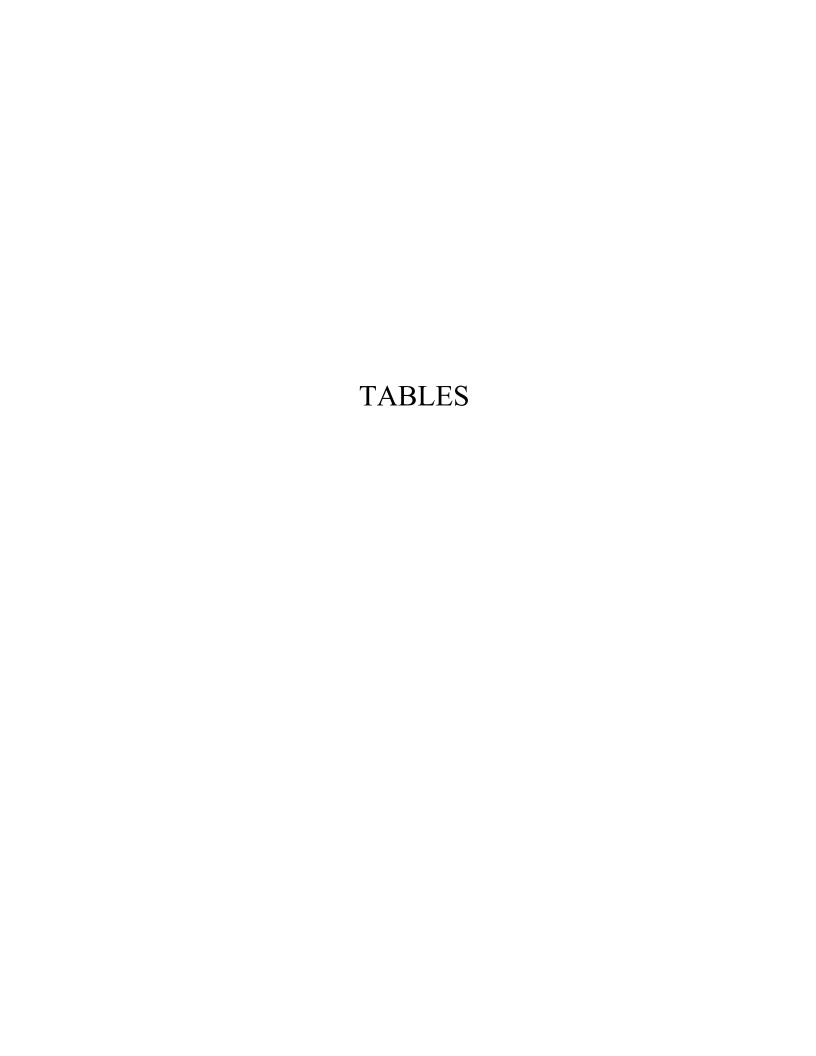


Table 1: Summary of Key Cobalt Analytical Data East Bottom Ash Pond - H.W. Pirkey Plant

Sample	Sample Date	Unit	Cobalt Concentration
Bottom Ash (Solid Material)	2/11/2019	mg/kg	6.1
SPLP Leachate of Bottom Ash	2/11/2019	mg/L	< 0.01
EBAP Pond Water	12/15/2018	mg/L	0.0024 J
WBAP Pond Water	11/4/2020	mg/L	0.000501
AD-2 - Average	May 2016 - November 2020	mg/L	0.0132
AD-31 - Average	May 2016 - November 2020	mg/L	0.0127
AD-32 - Average	May 2016 - November 2020	mg/L	0.0441

Notes:

mg/kg - milligram per kilogram

mg/L - milligram per liter

J - Estimated value. Result is less than the reporting limit but greater than or equal to the method detection limit.

A sample was collected from the WBAP on 11/4/2020 as a surrogate for the EBAP, as the EBAP did not contain free water. The same process water is stored in both the WBAP or EBAP.

Average values were calculated using all cobalt data collected under 40 CFR 257 Subpart D, excluding any identified outliers.

Table 2: Soil Cobalt Data East Bottom Ash Pond - H.W. Pirkey Plant

Location ID	Location	Sample Depth (ft bgs)	Cobalt (mg/kg)				
	Bulk	Soil Samples	\				
AD 2	EDAD Notrecula	25-27	9.45				
AD-2	EDAP Network	31-33	19.2				
AD-18	ERAP Network	8	3.60				
AD-10	EDAI Network	22	2.90				
ΔD-31	FRAP Network	12	1.90				
AD-31	EDAI NELWOIK	26	0.83				
ΔD-32	FRAP Network	11	1.70				
AD-32	EDAI NELWOIK	20-25	9.10				
		15	< 1.0				
AD-41	Upgradient	35	23.5				
		95	1.90				
		10	2.36				
		16	3.62				
B-2	Upgradient	71	10.30				
		82	0.83 1.70 9.10 < 1.0 23.5 1.90 2.36 3.62 10.30 7.21 3.11 1.30				
		87					
		10	1.30				
B-3	Upgradient	20	0.59				
		97	1.11				
	Solid Material I	Retained After Filtration					
AD-32	EBAP Network	13-33	5.4				
AD-18 EBAP Network 31-33 19.2 AD-18 EBAP Network 22 2.90 AD-31 EBAP Network 26 0.83 AD-32 EBAP Network 21 1.70 AD-41 Upgradient 35 23.5 B-2 Upgradient 71 10.30 B-3 Upgradient 20 0.59 B-3 Upgradient 20 0.59 Solid Material Retained After Filtration							
R-3	Ungradient	29-34	12.0				
D - 5	Opgradient	VAP 40-45	18.0				

Notes:

mg/kg- milligram per kilogram

ft bgs - feet below ground surface

For AD-XX locations, samples were collected from additional boreholes advanced in the immediate area of the location identified by the well ID. Samples were not collected from the cuttings of the borings advanced for well installation. Samples for B-2 and B-3 locations were collected from cores removed from the borehole during well lithology logging.

Depths for samples collected after filtration represent the screened interval for the permanent well where the sample was collected.

Table 3: X-Ray Diffraction Results East Bottom Ash Pond - H. W. Pirkey Plant

Constituent	VAP-B3-(40-45)
Quartz	15
Plagioclase Feldspar	0.5
Orthoclase	ND
Calcite	ND
Dolomite	ND
Siderite	0.5
Goethite	ND
Hematite	2
Pyrite	3
Kaolinite	42
Chlorite	4
Illite/Mica	6
Smectite	12
Amorphous	15

Notes:

ND: Not detected

VAP-B3-(40-45) is the centrifuged solid material from the groundwater sample collected at that interval.

Table 4: Summary of Key Lithium Analytical Data East Bottom Ash Pond - H.W. Pirkey Plant

Sample	Sample Date	Unit	Lithium Concentration
Bottom Ash (Solid Material)	2/11/2019	mg/kg	0.82 J
SPLP Leachate of Bottom Ash	2/11/2019	mg/L	0.011 J
EBAP Pond Water	12/15/2018	mg/L	0.023 J
WBAP Pond Water	11/4/2020	mg/L	0.0274
AD-31 - Average	May 2016 - November 2020	mg/L	0.0855
AD-32 - Average	May 2016 - November 2020	mg/L	0.139

Notes:

mg/kg - milligram per kilogram

mg/L - milligram per liter

J - Estimated value. Result is less than the reporting limit but greater than or equal to the method detection limit.

A sample was collected from the WBAP on 11/4/2020 as a surrogate for the EBAP, as the EBAP did not contain free water. The same process water is stored in both the WBAP or EBAP.

Average values were calculated using all lithium data collected under 40 CFR 257 Subpart D, excluding any identified outliers.

Table 5: Soil Lithium Data East Bottom Ash Pond - H.W. Pirkey Plant

Location ID	Sample Depth (ft bgs)	Lithium (mg/kg)			
	Bulk Soil Samples				
AD-32 11 0.53					
Content of the property of t	1.60				
	10	5.30			
D 2	16	3.97			
D-2	71	7.42			
	87	13.10			
	10	3.64			
B-3	20	2.59			
	97	0.53 1.60 5.30 3.97 7.42 13.10 3.64 2.59 11.10 2.9 J			
Lignite	N/A	2.9 J			
Solid	d Material Retained After Filts	ration			
AD-32	13-33	9.8 J			
B-2	38-48	6.5 J			
D 2	29-34	7.8 J			
B-3	VAP 40-45	13.0			

Notes:

J - estimated value

mg/kg- milligram per kilogram

ft bgs - feet below ground surface

For AD-32, samples were collected from additional boreholes advanced in the immediate area of the location identified by the well ID. Samples were not collected from the cuttings of the borings advanced for well installation. Samples for B-X locations were collected from cores removed from the borehole during well lithology logging.

Depths for samples collected after filtration represent the screened interval for the permanent well where the sample was collected.

VAP - vertical aquifer profiling

Table 6: Calculated Site-Specific Partition Coefficients East Bottom Ash Pond - H. W. Pirkey Plant

Source		B-2		Literature Value
Unit	mg/L mg/kg L/kg L/kg Aqueous			
Element	Aqueous Phase	Adsorbed	Kd	Kd
Li	0.081	6.5	80	43-370
K	2.6	1100	423	42-1200
Na	14	130	9	5.2-82

Source		B-3		Literature Value
Unit	Unit mg/L mg/kg L/kg L/kg Element Aqueous Phase Adsorbed Kd Kd Li 0.097 7.8 80 43-370 K 2.9 1100 379 42-1200	L/kg		
Element		Adsorbed	Kd	Kd
Li	0.097	7.8	80	43-370
K	2.9	1100	379	42-1200
Na	32	240	8	5.2-82

Source		AD-32		Literature Value
Unit mg/L mg/kg L/kg L/kg Element Aqueous Phase Adsorbed Kd Kd Li 0.11 9.8 89 43-370 K 3.9 1800 462 42-1200	mg/L mg/kg L/kg			
Element		Adsorbed	Kd	Kd
Li	0.11	9.8	89	43-370
K	3.9	1800	462	42-1200
Na	57	220	4	5.2-82

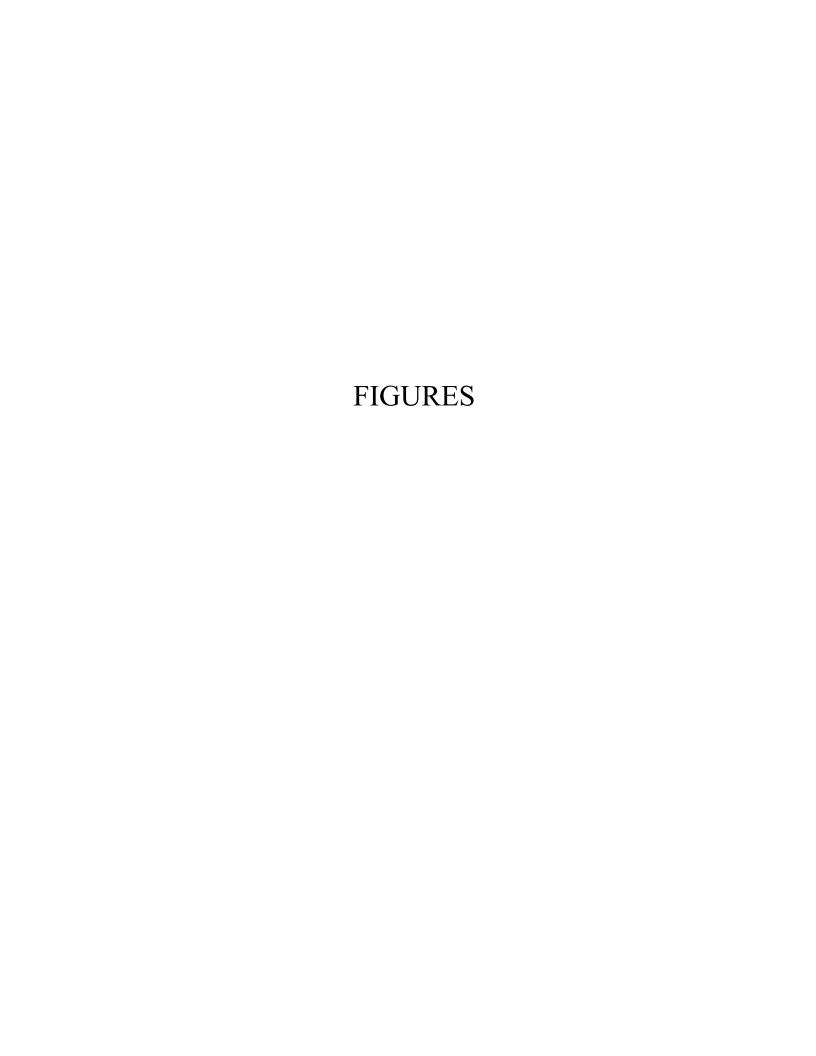
Notes:

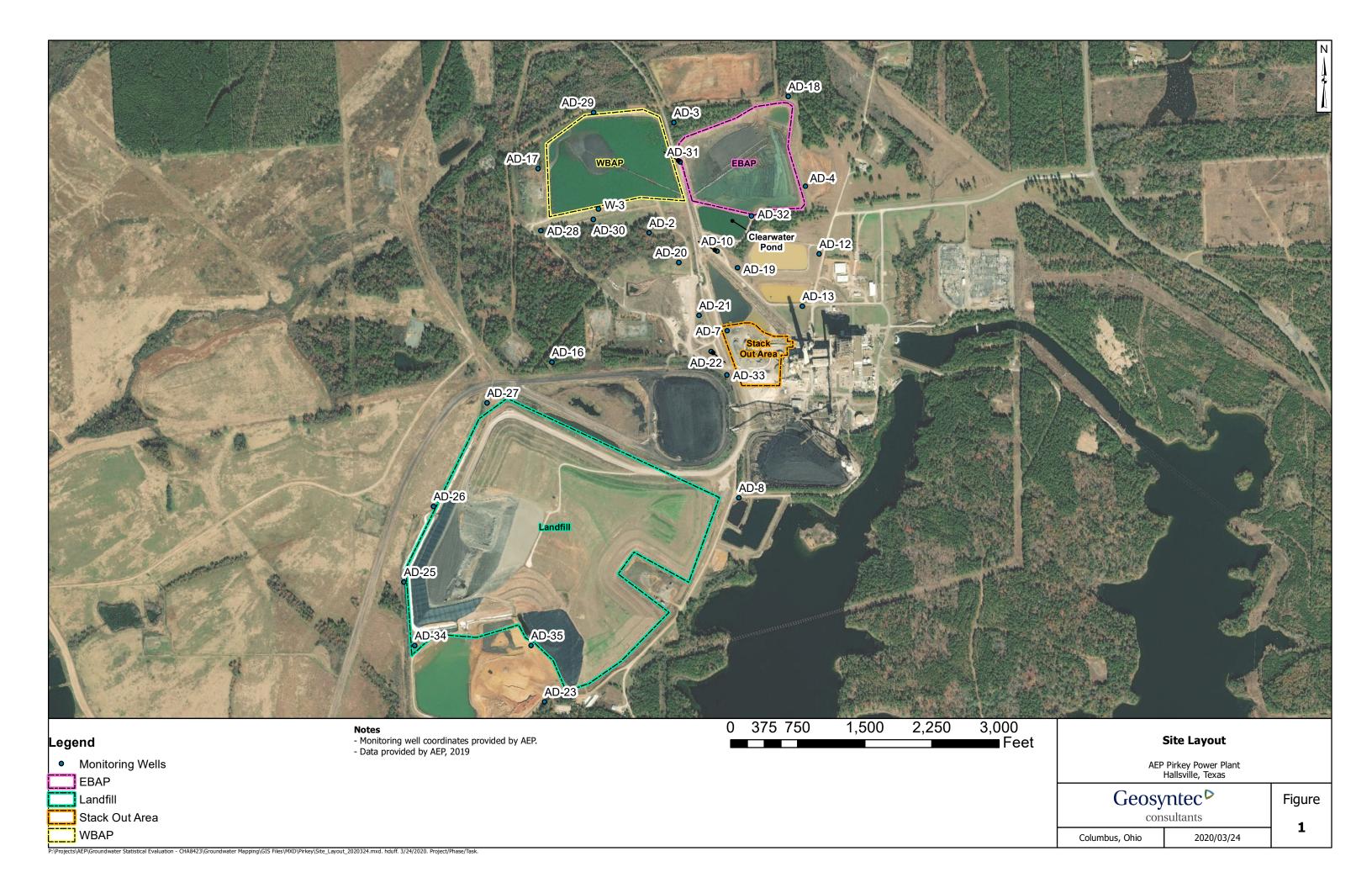
mg/L: milligrams per liter mg/kg: milligrams per kilogram

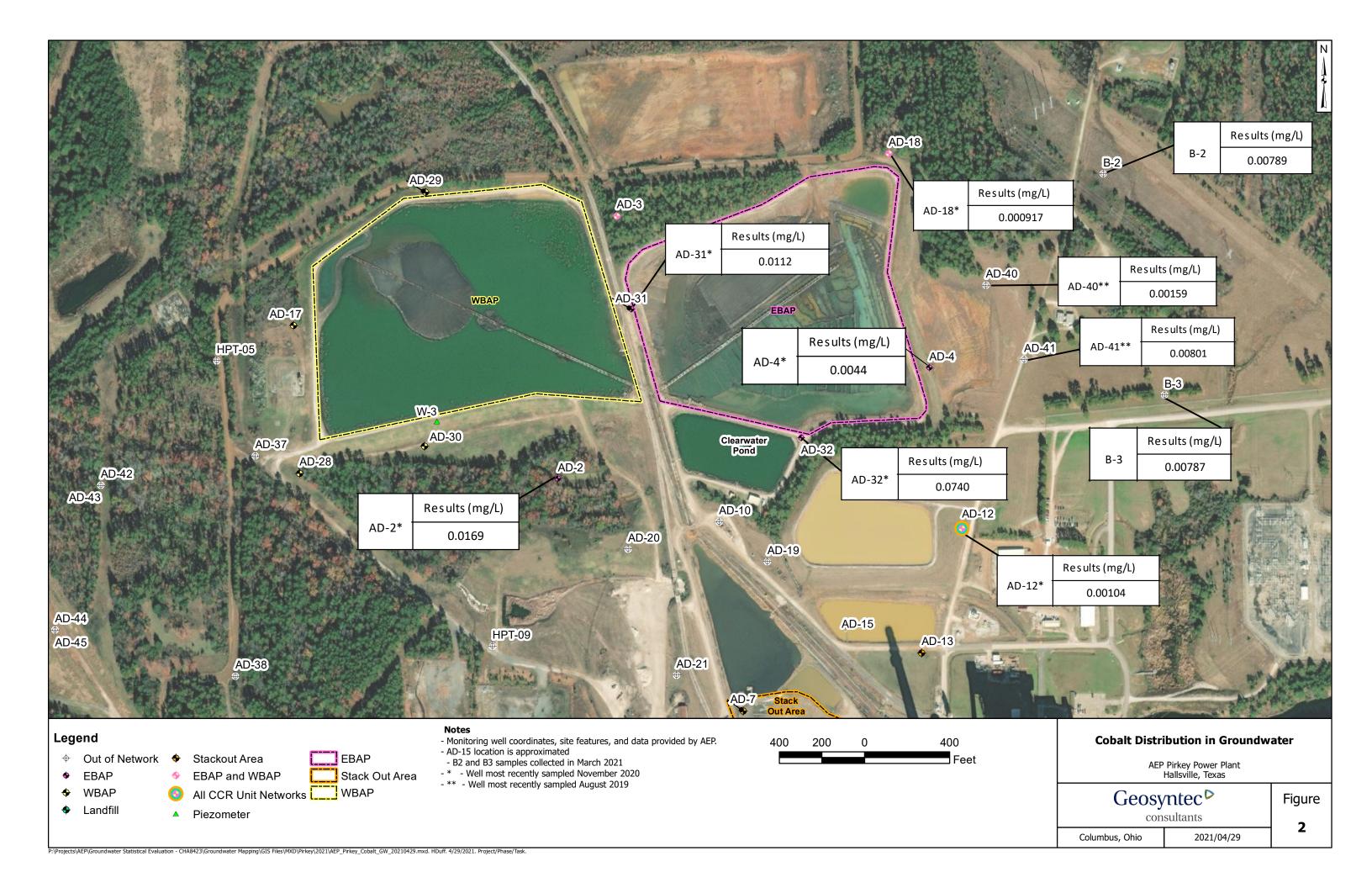
L/kg: liters per kilogram Kd: partition coefficient

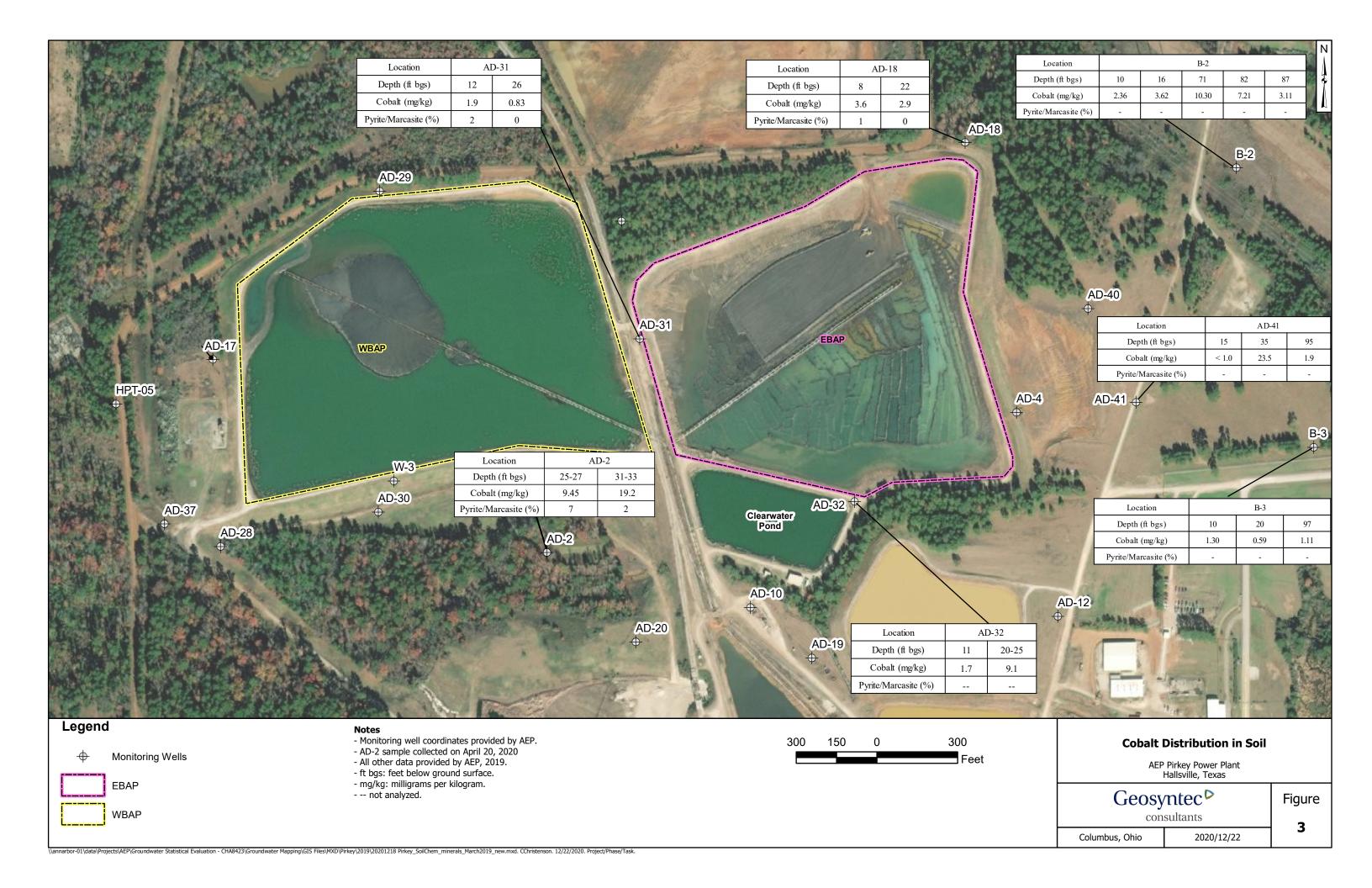
Adsorbed values are total metals concentrations reported by USEPA Method 6010B.

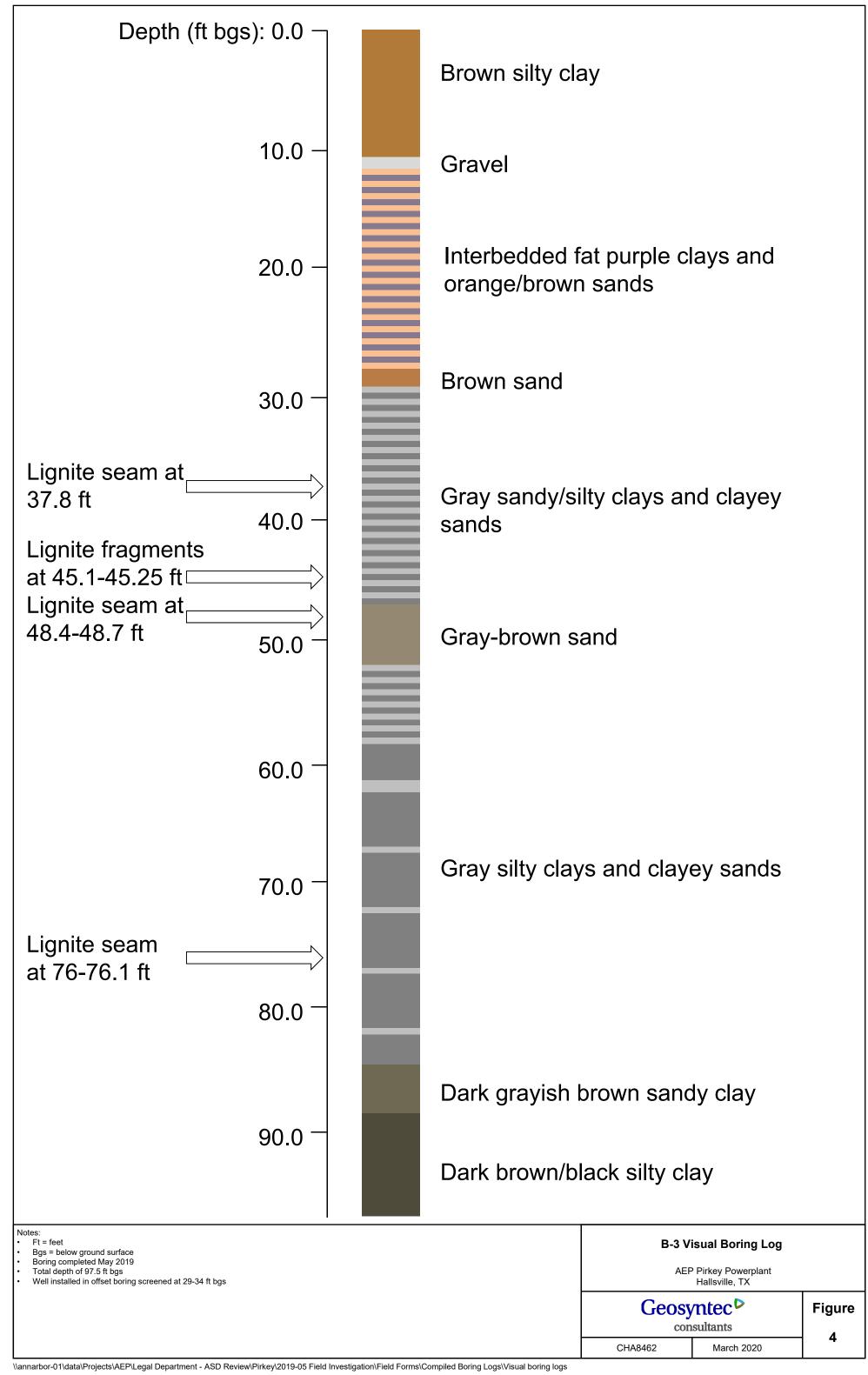
Literature values represent maximum and minimum values for the parameter as reported in Sheppard et al, 2009 (Table 4-1, all sites) and Sheppard et al, 2011 (Table 3-3 cultivated peat and wetland peat only).

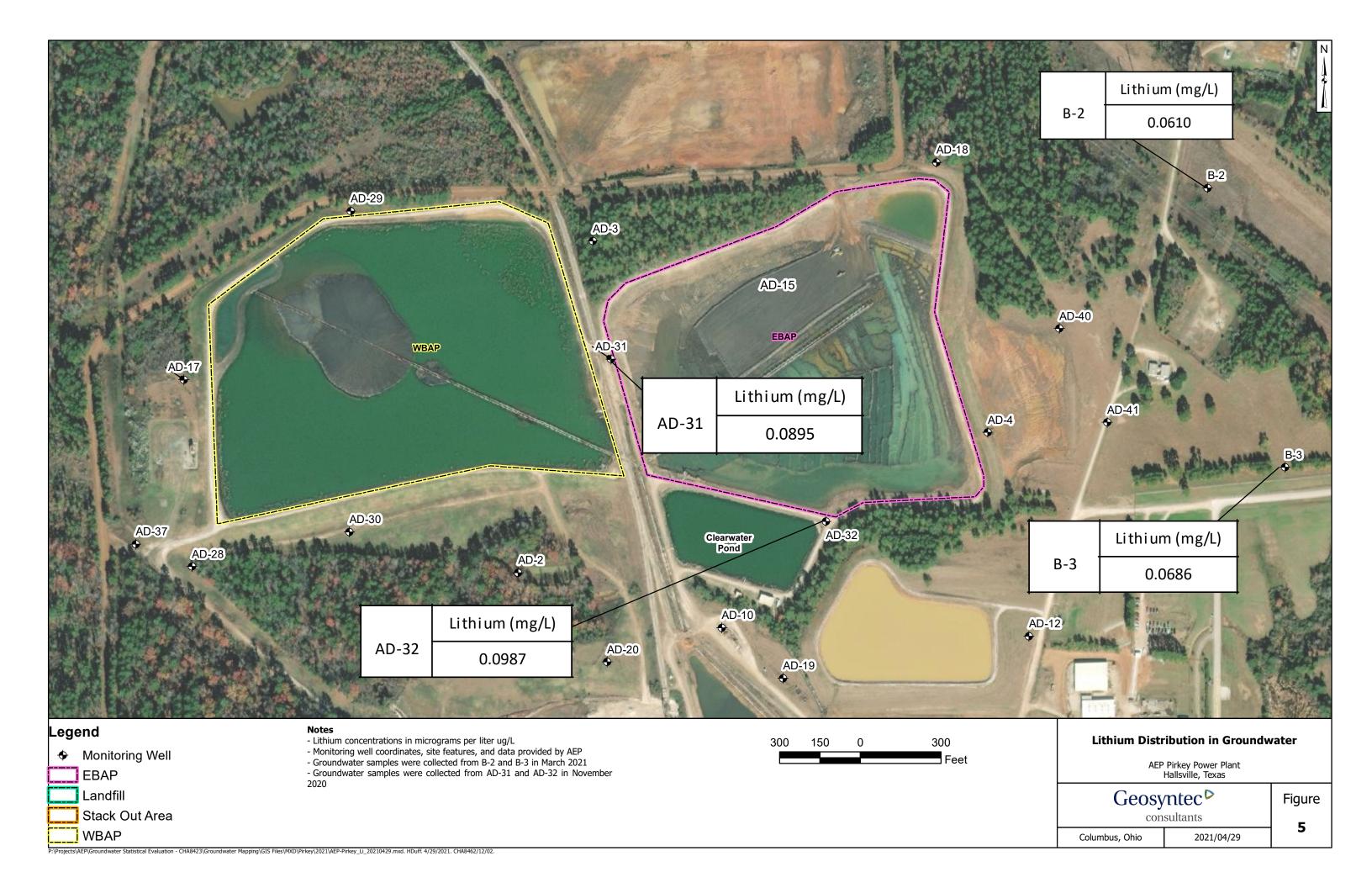


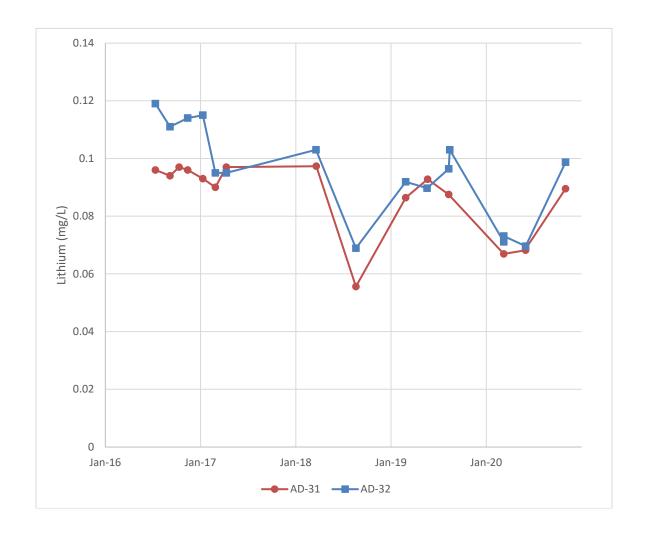












Notes: Lithium concentrations are shown milligrams per liter (mg/L). An outlier lithium value reported for AD-32 on 10/12/16 was excluded from the time series graph.

Lithium Time Series Graph Pirkey East Bottom Ash Pond

Figure

6



ATTACHMENT A SB-2 Boring Log

	ECT NOTION	An	,-2	/MW-2		ROJ.	Part	ELEV.			TE	4	58-2 1/20/20)
VLo V Lo L	loose led. Dense Dense	0- 4 4-10 10-30 30-50 >50	So. Msi St VSi		PENETROMET 0 - 0.25 0.25 - 0.5 0.5 - 1.0 1.0 - 2.0	CLAYS CER N - VA CER 2 - 4 - 8 - 15 -	LUE Li Dk. 4 G 8 T 15 R	COLORS LightBrBrown DarkBkBlack GreyBlBlue TanGrGrenn Red Y Yellow sh.Reddish.WhWhite	MATERIALS CI Clay, Clayer Si Silt, Silty Sa Sand, Sandy Ls Limestone Gr Gravel SiS Siltstone SS Sandstone Sh Shale, Shale,		SAN FF	Aedium Coarse	CHARAC Calc C	alcareous ignite irganic aminate lickensideo lightly eam (s)
100	9	T .	S			STRAT	UM DES	CRIPTION			ANDA	ARD	7	
Se whole Inte	Recovery	DEPTH F	SAMPLE	CONDITION OR CONSISTENCY	COLOR	MATERIALS OR ADJECTIVES	PREDOMINATE MATERIAL	CHARACTE OR MODIFICA		SEAT - 6"	1st - 6"	2nd - 6"	UNIFIED SOIL CLASSIFICATION	N - VALUE OR HAND
0-5	2' Rec	0		0-81	Br. H. Rd Br	Si	Sa	Silty Sand +	sace clay,					
5-10	2.5' Rec		-	1	Lt. Rd. Br			track root hairs	1 1 1 1/1				moist	10.5
2-10	2.0180		+		TI'KI'DI			- thin lenses (less than 1/4"	11.10			MOIST	(6-10
10-15	4' Rec	· Ø.		8-148	Lt. Rd St. Fd	SUSI	CI	Clay-som	Edind and s	1/4			moist	10-
			-		Br, Gray	julian.		clayer san	The state of the s	ede	1			- 9/2
			+	,				And in case of the last of the	race iron one	9/10	112	,51		
15-20	2'Rec	146	1	145	RLAN YILW.	Si,a	50	110000	some sand	Ela	3		VVMBIS	tto
			-	391	Br. Gray			and ironst		65	11		moist	(15
28-25	* No Re	6.	1		11/2 11	-	> (?	- centralet say	id seams in	51	14)	V. More	-(20-
25-30	2.5 R	Œ			Gray - DKG	ray ~		-gravel tremen	to saw sa	ne	25	16	1) sat, 9	125'-
		-	H		DK. BL	9/)	B	- convented au	particular production of the last of the l	æn	iews	00	110/24	nr.
					(24-5)	0			sande 25	10	Alla.	12"	- MOIST	27.
0. 25	2/8							e 27/1						
30-35	3'Rec		1		Cotta EX	CYCL.	•	- sat silty sa			111	-	Sat (33.5
					- 17			* some u.f. a	VDCUM ON	ctal	Sin	de	exsand	32.
25110	4' REC	00	H	-9 11:	1164	- 0)(01	* some u.f. a	of sand sea	NUS	(25	-40	y v,n	16154
70-40	TRA	21		31-70	4. Gray, 6	vay Up	Si	Chayey Sandy	Soft sola	,0	391	445	MAZTICA	- (29
		-				100		B.T. CHO!	1				Monto	(31)
												1	42%	
								#25.27			5			4
			H					*31-33'	sheated e 10	35			11.	-
			П											

* GPS: 32,46522, -94,49032 (12'E',)
3.5'N)
of AD-2/MW-2,

ATTACHMENT B SB-2 Boring Photographic Log

Geosyntec consultants

Client: AEP Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond Site Location: Hallsville, Texas

Photograph 1

Date: 4/21/2020

Direction: N/A

Comments:

0-5 foot interval of SB-2.



Photograph 2

Date: 4/21/2020

Direction: N/A

Comments:

5-10 foot interval of

SB-2.



1

Geosyntec consultants

Client: AEP Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond Site Location: Hallsville, Texas

Photograph 3

Date: 4/21/2020

Direction: N/A

Comments:

10-15 foot interval of

SB-2.



Photograph 4

Date: 4/21/2020

Direction: N/A

Comments:

15-20 foot interval of SB-2. Recovery of this interval was limited.



Geosyntec consultants

20.12.22

Client: AEP Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond Site Location: Hallsville, Texas

Photograph 5

Date: 4/21/2020

Direction: N/A

Comments:

20-25 foot interval of SB-2. Recovery of this interval was limited.



Photograph 6

Date: 4/21/2020

Direction: N/A

Comments:

25-30 foot interval of SB-2. Very little of this interval was recovered. A color change was observed from red to dark brown/black. A sample was collected from this interval.



Geosyntec consultants

Client: AEP Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond Site Location: Hallsville, Texas

Photograph 9

Date: 4/21/2020

Direction: N/A

Comments:

30-35 foot interval of SB-2. Very little of this interval was recovered.. A sample was collected from this interval.



Photograph 10

Date: 4/21/2020

Direction: N/A

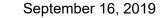
Comments:

35-40 foot interval of

SB-2



ATTACHMENT C SEM/EDS Analysis



via Email: BSass@geosyntec.com

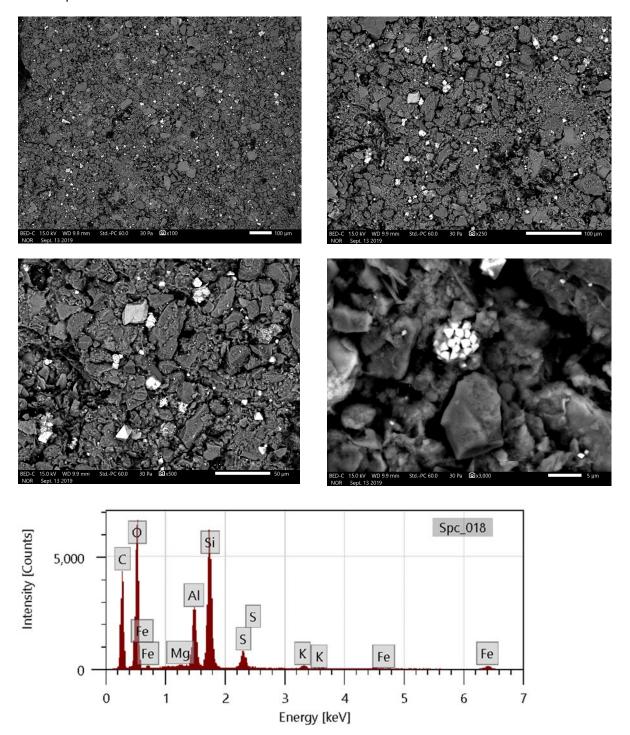


Dr. Bruce Sass 941 Chatham Lane, Suite 103, Columbus, OH 43221

Spc_004 Intensity [Counts] 1,500 Αl 1,000 500 Fe

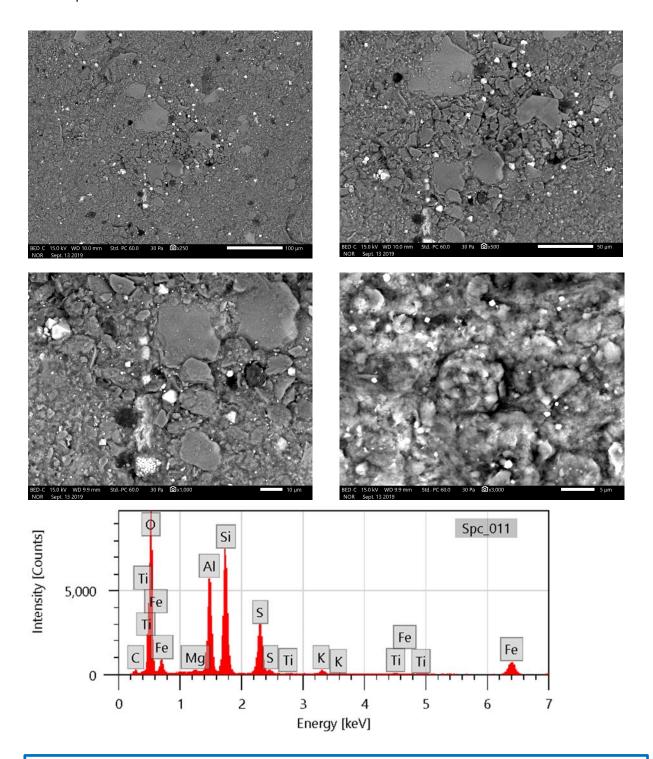
Lignite. Backscattered electron micrographs show the sample at 100X, 1,100X, and 1,500X. EDS spectrum at bottom is an area scan of the region shown in top right micrograph. Bright particles are mostly quartz and feldspar. Major peaks for carbon, oxygen, silicon, and aluminum suggest coal and clay.

Energy [keV]



Sample VAP B3 40-45. Backscattered electron micrographs show the sample at 100X, 250X, 500X, and 3000X. EDS spectrum at bottom is an area scan of the region shown at 500X. Bright particles are pyrite (framboid in bottom right micrograph). Major peaks for carbon, oxygen, silicon, and aluminum suggest coal and clay.





Sample VAP B3 50-55. Backscattered electron micrographs show the sample at 250X, 500X, 1000X, and 3000X. EDS spectrum at bottom is an area scan of the region shown at 3000X. Bright particles are mostly pyrite (framboid in bottom left micrograph); occasional particles of Fe-Ti oxide are detected. Major peaks for oxygen, silicon, and aluminum suggest clay. Large blocky particles are mostly quartz, feldspar, and clay.



ATTACHMENT D Certification by Qualified Professional Engineer

CERTIFICATION BY A QUALIFIED PROFESSIONAL ENGINEER

I certify that the selected and above described alternative source demonstration is appropriate for evaluating the groundwater monitoring data for the Pirkey East Bottom Ash Pond CCR management area and that the requirements of 40 CFR 257.95(g)(3)(ii) have been met.

Beth Ann Gross

Printed Name of Licensed Professional Engineer

Signature

Geosyntec Consultants 2039 Centre Pointe Blvd, Suite 103 Tallahassee, Florida 32308

Texas Registered Engineering Firm No. F-1182

79864 License Number Texas
Licensing State

Date