ALTERNATIVE SOURCE DEMONSTRATION REPORT TEXAS STATE CCR RULE

H.W. Pirkey Power Plant East Bottom Ash Pond Hallsville, Texas

Submitted to



1 Riverside Plaza Columbus, Ohio 43215-2372

Submitted by



engineers | scientists | innovators

941 Chatham Lane, Suite 103 Columbus, OH 43221

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LIST OF ACRONYMS

- AEP American Electric Power
- ASD Alternative Source Demonstration
- CCR Coal Combustion Residuals
- EBAP East Bottom Ash Pond
- EDS Energy Dispersive Spectroscopic Analyzer
- EPRI Electric Power Research Institute
- GSC Groundwater Stats Consulting, LLC
- GWPS Groundwater Protection Standard
- LCL Lower Confidence Limit
- MCL Maximum Contaminant Level
- QA Quality Assurance
- QC Quality Control
- SEM Scanning Electron Microscopy
- SPLP Synthetic Precipitation Leaching Profile
- SSL Statistically Significant Level
- TAC Texas Administrative Code
- TCEQ Texas Commission on Environmental Quality
- UTL Upper Tolerance Limit
- USEPA United States Environmental Protection Agency
- VAP Vertical Aquifer Profiling
- WBAP West Bottom Ash Pond
- XRD X-Ray Diffraction

SECTION 1

INTRODUCTION AND SUMMARY

This Alternative Source Demonstration (ASD) report has been prepared to address statistically significant levels (SSLs) for cobalt and lithium in the groundwater monitoring network at the H.W. Pirkey Plant East Bottom Ash Pond (EBAP), located in Hallsville, Texas, following the second semi-annual assessment monitoring event of 2021.

The H.W. Pirkey Plant has four coal combustion residuals (CCR) storage units regulated by the Texas Commission on Environmental Quality (TCEQ) under Registration No. CCR104, including the EBAP (**Figure 1**). The EBAP is also registered as a surface impoundment under TCEQ Industrial and Hazardous Waste Solid Waste Registration No. 33240. In November 2021, a semi-annual assessment monitoring event was conducted at the EBAP in accordance with 30 TAC §352.951(a). The monitoring data were submitted to Groundwater Stats Consulting, LLC (GSC) for statistical analysis. Groundwater protection standards (GWPSs) were established for each Appendix IV parameter in accordance with the statistical analysis plan developed for the unit (Geosyntec, 2020a) and the United States Environmental Protection Agency's (USEPA's) *Statistical Analysis of Groundwater Monitoring Data at RCRA Facilities – Unified Guidance* (Unified Guidance; USEPA, 2009). The GWPS for each parameter was established as the greater of either the background concentration or, for constituents with a maximum contaminant level (MCL), the MCL. To determine background concentrations, an upper tolerance limit (UTL) was calculated using pooled data from the background wells collected during the background monitoring and assessment monitoring events.

Confidence intervals were re-calculated for the Appendix IV parameters at the compliance wells to assess whether these parameters were present at a statistically significant level (SSL) above the GWPSs. An SSL was concluded if the lower confidence limit (LCL) of a parameter exceeded the GWPS (i.e., if the entire confidence interval exceeded the GWPS). The following SSLs were identified at the Pirkey EBAP (Geosyntec, 2022):

- The LCLs for cobalt exceeded the GWPS of 0.0094 mg/L at AD-2 (0.0100 mg/L), AD-31 (0.00956 mg/L), and AD-32 (0.0250 mg/L).
- The LCL for lithium exceeded the GWPS of 0.0550 mg/L at AD-31 (0.0664 mg/L) and AD-32 (0.0781 mg/L).

No other SSLs were identified.

1.1 <u>CCR Rule Requirements</u>

TCEQ regulations regarding assessment monitoring programs for CCR landfills and surface impoundments (TCEQ, 2020a) provide owners and operators with the option to make an ASD when an SSL is identified (30 TAC §352.951(e)):

... In making a demonstration under this subsection, the owner or operator must, within 90 days of detecting a statistically significant level above the groundwater protection standard of any constituent listed in Appendix IV adopted by reference in §352.1431 of this title, submit a report prepared and certified in accordance with §352.4 of this title (relating to Engineering and Geoscientific Information) to the executive director, and any local pollution agency with jurisdiction that has requested to be notified, demonstrating that a source other than a CCR unit caused the exceedance or that the exceedance resulted from error in sampling, analysis, statistical evaluation, or natural variation in groundwater quality.

Pursuant to 30 TAC §352.951(e), Geosyntec Consultants, Inc. (Geosyntec) has prepared this ASD report to document that the SSLs identified for cobalt and lithium are from a source other than the EBAP.

1.2 Demonstration of Alternative Sources

An evaluation was completed to assess possible alternative sources to which the identified SSLs could be attributed. Alternative sources were identified amongst five types, based on methodology provided by EPRI (2017):

- ASD Type I: Sampling Causes;
- ASD Type II: Laboratory Causes;
- ASD Type III: Statistical Evaluation Causes;
- ASD Type IV: Natural Variation; and
- ASD Type V: Alternative Sources.

A demonstration was conducted to show that the SSLs identified for cobalt and lithium were based on a Type IV cause and not by a release from the Pirkey EBAP.

SECTION 2

ALTERNATIVE SOURCE DEMONSTRATION

The TCEQ CCR Rule allows the owner or operator 90 days from the determination of an SSL to demonstrate that a source other than the CCR unit caused the SSL. Descriptions of the regional geology and site hydrogeology and the methodology used to evaluate the SSLs and the proposed alternative source are described below.

2.1 <u>Regional Geology/Site Hydrogeology</u>

The EBAP is positioned on an outcrop of the Eocene-age Recklaw Formation, which consists predominantly of clay and fine-grained sand (Arcadis, 2016). The Recklaw Formation is underlain by the Carrizo Sand, which crops out in the topographically lower southern portion of the plant. The Carrizo Sand consists of fine to medium grained sand interbedded with silt and clay.

The EBAP monitoring well network monitors groundwater within the Uppermost Aquifer, which was defined by Arcadis (2016) as very fine to fine grained clayey and silty sand with an average thickness of approximately 15 feet. Geologic cross-section A-A' from the EBAP Groundwater Monitoring Well Network Report (Arcadis, 2016) shows the subsurface geometry of the Uppermost Aquifer (indicated on the figure as clayey silty sand, tan to gray) underlying the EBAP and the West Bottom Ash Pond (WBAP). This figure is provided as **Attachment A**. **Attachment A** demonstrates lateral continuity of the Uppermost Aquifer spanning the entire length of the EBAP.

Groundwater flow direction in the area of the EBAP is west-southwesterly (**Figure 1**). Seasonal variability in groundwater flow has not been observed since the monitoring well network was installed. Groundwater flow through the Uppermost Aquifer contains a hydraulic gradient of approximately 0.01 feet per foot. The EBAP monitoring well network consists of upgradient monitoring wells AD-4, AD-12, and AD-18, and compliance wells AD-2, AD-3, AD-31, and AD-32, all of which are screened within the Uppermost Aquifer.

2.2 <u>Proposed Alternative Source</u>

An initial review of site geochemistry, site historical data, and laboratory quality assurance/quality control (QA/QC) data did not identify alternative sources for cobalt and lithium due to Type I (sampling), Type II (laboratory), Type III (statistical evaluation), or Type V (anthropologic) issues. Groundwater sampling, laboratory analysis, and statistical evaluations were generally completed in accordance with 30 TAC §352.931 and the draft TCEQ guidance for groundwater monitoring (TCEQ, 2020b). As described below, the SSLs have been attributed to natural variation associated with the underlying geology, which is a Type IV (natural variation) issue.

2.2.1 Cobalt

Previous ASDs for cobalt at the EBAP provided evidence that cobalt is present in the aquifer geologic media at the site and that the observed cobalt concentrations were due to natural variation (Geosyntec, 2019a; Geosyntec, 2019b; Geosyntec, 2020b; Geosyntec, 2020c; Geosyntec, 2021a; Geosyntec, 2021b) of native geogenic sources. The previous ASDs demonstrated how the EBAP was not a source for cobalt in downgradient groundwater, based on observed concentrations of cobalt both in the ash material and in leachate from Synthetic Precipitation Leaching Procedure (SPLP) analysis (SW-846 Test Method 1312, [USEPA, 1994]) of the ash material. Cobalt was not detected in the SPLP ash leachate above the reporting limit of 0.01 mg/L, which is lower than the average concentrations observed at the wells of interest (**Table 1**).

Surface water samples were collected from the EBAP and West Bottom Ash Pond (WBAP) to characterize the total cobalt concentrations. Cobalt was detected in a sample collected on June 2, 2020 from the EBAP at an estimated concentration of 0.000080 mg/L (**Table 1**). Sampling of the EBAP was attempted again in November 2020 but was unsuccessful as the EBAP did not contain free water at the time of the sampling event. A sample was collected from the WBAP as a surrogate for the EBAP sample. Cobalt was detected at a concentration of 0.000501 mg/L in this WBAP surrogate sample (**Table 1**). The EBAP and WBAP receive the same process water, with the use of each pond dependent on available freeboard and cleaning schedule; thus, there is a basis for the equivalency of these two surface water samples. No changes to material handling or plant operations have occurred which would change the anticipated cobalt concentrations in the ponds since these samples were collected. These concentrations are lower than all reported cobalt concentrations for in network wells from the most recent sampling event and over an order of magnitude lower than the average concentration in groundwater at the wells of interest (**Table 1**; **Figure 2**). Thus, the EBAP is not the likely source of cobalt at AD-2, AD-31, and AD-32.

As noted in the previous ASDs, soil samples collected across the site, including from locations near the EBAP, identified cobalt in the aquifer solids at varying concentrations. SB-2 was advanced in the vicinity of AD-2 in April 2020 to re-log the geology at AD-2 and collect samples for laboratory analysis of total metals and mineralogy. The SB-2 field boring log, which was generated by Auckland Consulting LLC, is provided as **Attachment B**. Cobalt was detected at SB-2 at concentrations of 9.45 milligrams per kilogram (mg/kg) at 25-27 feet below ground surface (bgs) and 19.2 mg/kg at 31-33 feet bgs (**Table 2**). These cobalt concentrations are greater than the concentration of cobalt present in the bottom ash (**Table 1**). Both samples correlate to the depth of the monitoring well screen of AD-2 (20-40 feet bgs), indicating that naturally occurring cobalt is present in aquifer solids within the AD-2 screened interval. Cobalt was also identified in the aquifer solids at varying concentrations at other locations throughout the site, with the highest value of 23.5 mg/kg reported at AD-41, which is upgradient of the EBAP (**Figure 3**).

In addition to the analysis of total cobalt, soil samples were submitted for mineralogical analysis to determine the mineral composition of soils near the EBAP. X-ray diffraction (XRD) analysis of soils from SB-2 identified pyrite (an iron sulfide) in samples collected at 25-27 feet bgs and 31-33 feet bgs at concentrations up to 7% by weight (**Figure 3**). Cobalt is known to undergo isomorphic

substitution for iron in crystalline iron minerals such as pyrite due to their similar ionic radii of approximately 1.56 angstroms (Å) for iron vs. 1.52 Å for cobalt (Clementi and Raimondi, 1963; Krupka and Serne, 2002; Hitzman et al., 2017). The presence of iron-bearing minerals in soil near the EBAP constitutes a potential source of naturally occurring cobalt.

The aquifer solids at SB-2 are distinctly red in color at shallow depths, as illustrated in the photolog of soil cores provided in **Attachment C**. While shallow samples were not collected for mineralogical analysis, red color in soils is often associated with the presence of oxidized ironbearing minerals such as hematite and goethite. The red color of the soil suggests the presence of iron oxide and hydroxide minerals within the shallow depth interval. The alteration of pyrite to these iron oxide and hydroxide minerals under oxidizing conditions is also a well-understood phenomenon, including in formations in east Texas (Senkayi et al., 1986; Dixon et al., 1982). It is likely that the pyrite alteration process is resulting in the release of isomorphically substituted cobalt from the pyrite crystal structure as it undergoes oxidative transformation to iron oxide/hydroxide minerals.

As described in the previous ASDs, vertical aquifer profiling (VAP) was used to collect groundwater samples from upgradient locations B-2 and B-3 during the soil boring and sample collection process (Geosyntec, 2019b). A groundwater sample was also collected from AD-32, an existing well within the EBAP groundwater monitoring network. Solid phases within these groundwater samples were separated and submitted for analysis of chemical composition. For the VAP samples, separation was completed using a centrifuge due to the high abundance of suspended solids. For the groundwater sample at AD-32, the sample was filtered using a 1.5-micron filter. Based on total metals analysis, cobalt was identified both in the centrifuged solid material collected from upgradient VAP location B-3 [VAP-B3-(40-45)] and in the material retained on the filter after processing groundwater from permanent monitoring wells B-2 and B-3 (**Table 2**). The concentrations of cobalt in the solid material retained after filtration were comparable to the bulk soil samples collected from the same locations.

The solid sample [VAP-B3-(40-45)] was submitted for mineralogical analysis via XRD and scanning electron microscopy (SEM) using an energy dispersive spectroscopic analyzer (EDS). The XRD results identified pyrite as approximately 3% of the solid phase (**Table 3**). Pyrite was identified during SEM/EDS analysis of lignite which is mined immediately adjacent to the site. Logging completed while the VAP boring was advanced identified coal at several intervals, including 45 and 48 feet bgs (**Figure 4**). Furthermore, SEM/EDS of both centrifuged solid samples [VAP-B3-(40-45) and VAP-B3-(50-55)] identified pyrite in backscattered electron micrographs by the distinctive framboidal morphology (Harris et al., 1981; Sawlowicz, 2000). Major peaks representing iron and sulfur were identified in the EDS spectrum, which further support the identification of pyrite (**Attachment D**). While cobalt was not identified in the EDS spectrum, it is likely present at concentrations below the detection limit.

The EBAP was not identified as the source of cobalt at wells in the EBAP network based on the low concentrations of cobalt in the pond itself. Cobalt in the EBAP network groundwater is believed to be a result of natural variability within the aquifer. Naturally occurring cobalt is known

to substitute for iron in iron-bearing minerals. The presence of iron sulfide pyrite and iron oxides/hydroxides hematite and goethite have been confirmed at AD-2 and across the Site. The weathering of pyritic minerals to iron oxide/hydroxide minerals may be resulting in the release of cobalt into groundwater from the crystal structure of these aquifer minerals.

2.2.2 Lithium

Previous ASDs for lithium at the EBAP attributed the observed lithium exceedances to variations in lithium associated with the suspended native aquifer solids that likely originate from naturally occurring lignite present in these soils. These native lithium-containing aquifer solids are ubiquitous in the aquifer based on the presence of lithium at upgradient locations and in the solid phase (Geosyntec, 2019b; Geosyntec, 2020b; Geosyntec, 2020c; Geosyntec, 2021a; Geosyntec, 2021b). Data gathered in support of the prior ASDs and recent results provide additional evidence that the observed lithium concentrations at AD-31 and AD-32 are naturally occurring and are due to natural variation in the aquifer (Type IV ASD).

As discussed in Section 2.1.1, surface water samples were collected directly from the EBAP and WBAP. Lithium was detected in the June 2, 2020 EBAP sample at a concentration of 0.0295 mg/L, which is comparable to the concentration of 0.0274 mg/L reported for the WBAP water on November 4, 2020 (**Figure 5, Table 4**). The mobile fraction identified in the bottom ash by SPLP was even lower, with an estimated lithium concentration of 0.011 mg/L. These concentrations are lower than the average lithium concentrations at AD-31 (0.0824 mg/L) and AD-32 (0.0863 mg/L) (**Table 4**). Thus, the EBAP is not the source of lithium at AD-31 and AD-32.

Groundwater samples collected from upgradient wells B-2 and B-3 in November 2021 had total lithium concentrations of 0.0554 mg/L and 0.0871 mg/L, respectively; the reported concentration at B-3 is greater than both the GWPS of 0.0590 mg/L and the concentrations of lithium observed at AD-31 and AD-32 (**Figure 5**). Because B-2 and B-3 were installed at locations upgradient to and unimpacted by site activities, these lithium concentrations suggest that dissolved lithium is naturally present at concentrations above the GWPS across the site at variable concentrations, and not limited to AD-31 and AD-32. It is noted that B-2 and B-3 are not part of the monitoring network for the EBAP, and as such the lithium concentrations in groundwater from these wells are not considered in calculating the GWPS for the CCR unit.

As described in Section 2.1.1, groundwater samples were collected from B-2, B-3, and AD-32 and filtered to separate solids. Groundwater was also collected from a VAP boring (VAP-B3-(40-45)) and centrifuged to separate solids. Lithium was detected in the solid material separated from these groundwater samples at concentrations comparable to bulk soil at all locations, providing evidence that the particulates captured during groundwater sampling contain lithium (**Table 5**).

2.2.2.1 Calculated Partition Coefficients

A previous ASD for lithium at the EBAP discussed proposed lithium mobility in groundwater due to desorption from clay minerals associated with naturally occurring lignite material. This mechanism was posited as the source of lithium in both upgradient and downgradient wells at the EBAP (Geosyntec, 2019b). Previously completed XRD analysis of centrifuged solid material samples (VAP-B3-(40-45)) found that clay minerals, including kaolinite, smectite, and illite/mica, made up at least 60% of the aquifer solid (**Table 3**). SEM/EDS analysis also identified the presence of silicon, aluminum, and oxygen, all of which are components of clay minerals (**Attachment D**). The backscattered electron micrographs of these samples also identified clay particles by morphology. The largest clay particles (> 5 μ m) are likely kaolinite, while smectite and illite dominate the smaller size fraction. These clay minerals, particularly smectite and illite, are known to retain cations such as lithium via incorporation into the octahedral layer of the mineral structure and through cation exchange processes.

Mass measurements and total metal concentrations in the solid materials separated from the groundwater samples during filtration and the filtered groundwater concentrations were used to calculate partition coefficients values (K_d) for lithium, potassium, and sodium. Details about the K_d calculation are provided in the previous ASD (Geosyntec, 2019b). K_d values for groundwater and particulates collected from wells B-2, B-3, and AD-32 were comparable to literature K_d values reported for organic-rich media such as bogs and peat beds (Sheppard et al., 2009; Sheppard et al., 2011), providing further evidence that lithium mobility in site groundwater is similar to other sites with organic-rich soils (**Table 6**). Additionally, the calculated K_d values for Pirkey soils were consistent with the literature, with potassium having the highest K_d (greatest affinity for sorption) and sodium the lowest K_d (least affinity for sorption). Furthermore, the values are similar for groundwater from all three wells, suggesting a universal mechanism controlling lithium, sodium, and potassium mobility in groundwater. Since the site-specific Kd values were calculated, lithium concentrations at the wells of interest have remained consistent, suggesting that the clay mineralogy mechanism is still controlling lithium groundwater concentrations (**Figure 6**).

These multiple lines of evidence show that elevated lithium concentrations at AD-31 and AD-32 are not due to a release from the EBAP, and instead can be attributed to natural variation (Type IV ASD). This variation appears related to the distribution of clay fractions associated with lignite materials in the soil aquifer material.

2.3 <u>Sampling Requirements</u>

As the ASD presented above supports the position that the identified SSLs are not due to a release from the Pirkey EBAP, the unit will remain in the assessment monitoring program. Groundwater at the unit will continue to be sampled for Appendix IV parameters on a semiannual basis.

SECTION 3

CONCLUSIONS AND RECOMMENDATIONS

The preceding information serves as the ASD prepared in accordance with 30 TAC §352.951(e) and supports the position that the SSLs for cobalt and lithium identified during assessment monitoring in November 2021 were not due to a release from the EBAP. The identified SSLs should instead be attributed to natural variation in the underlying geology. Therefore, no further action is warranted, and the Pirkey EBAP will remain in the assessment monitoring program. Certification of this ASD by a qualified professional engineer is provided in **Attachment E**.

SECTION 4

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TABLES

Table 1: Summary of Key Cobalt Analytical DataEast Bottom Ash Pond - H.W. Pirkey Plant

Sample	Sample Date	Unit	Cobalt Concentration
Bottom Ash (Solid Material)	2/11/2019	mg/kg	6.1
SPLP Leachate of Bottom Ash	2/11/2019	mg/L	<0.01
EBAP Pond Water	6/2/2020	mg/L	0.000080
WBAP Pond Water	11/4/2020	mg/L	0.000501
AD-2 - Average	May 2016 - November 2021	mg/L	0.0149
AD-31 - Average	May 2016 - November 2021	mg/L	0.0121
AD-32 - Average	May 2016 - November 2021	mg/L	0.0450

Notes:

mg/kg - milligram per kilogram

mg/L - milligram per liter

J - Estimated value. Result is less than the reporting limit but greater than or equal to the method detection limit.

A sample was collected from the WBAP on 11/4/2020 as a surrogate for the EBAP, as the EBAP did not contain free water. The same process water is stored in both the WBAP and EBAP.

Average values were calculated using all cobalt data collected under 40 CFR 257 Subpart D, excluding any identified outliers.

Table 2: Soil Cobalt DataEast Bottom Ash Pond - H.W. Pirkey Plant

Location ID	Location	Sample Depth (ft bgs)	Cobalt (mg/kg)					
	Bulk Soil Samples							
AD-2	EBAP Network	25-27	9.45					
AD-2	EDAP Network	31-33	19.2					
AD-18	EBAP Network	8	3.60					
AD-18	EDAI Network	22	2.90					
AD-31	EBAP Network	12	1.90					
AD-31	EDAI Network	26	0.83					
AD-32	EBAP Network	11	1.70					
AD-32	EDAI Network	20-25	9.10					
	Upgradient	15	< 1.0					
AD-41		35	23.5					
		95	1.90					
		10	2.36					
		16	3.62					
B-2	Upgradient	71	10.30					
		82	7.21					
		87	3.11					
		10	1.30					
B-3	Upgradient	20	0.59					
		97	1.11					
	Solid Material Retained After Filtration							
AD-32	EBAP Network	13-33	5.4					
B-2	Upgradient	38-48	4.3					
B-3	Upgradient -	29-34	12.0					
D-2	Opgradient	VAP 40-45	18.0					

Notes:

mg/kg- milligram per kilogram

ft bgs - feet below ground surface

For AD-XX locations, samples were collected from additional boreholes advanced in the immediate area of the location identified by the well ID. Samples were not collected from the cuttings of the borings advanced for well installation. Samples for B-2 and B-3 locations were collected from cores removed from the borehole during well lithology logging.

Depths for samples collected after filtration represent the screened interval for the permanent well where the sample was collected.

Table 3: X-Ray Diffraction ResultsEast Bottom Ash Pond - H. W. Pirkey Plant

Constituent	VAP-B3-(40-45)
Quartz	15
Plagioclase Feldspar	0.5
Orthoclase	ND
Calcite	ND
Dolomite	ND
Siderite	0.5
Goethite	ND
Hematite	2
Pyrite	3
Kaolinite	42
Chlorite	4
Illite/Mica	6
Smectite	12
Amorphous	15

Notes:

Results given in units of relative % abundance ND: Not detected

VAP-B3-(40-45) is the centrifuged solid

material from the groundwater sample collected at that interval.

Table 4: Summary of Key Lithium Analytical DataEast Bottom Ash Pond - H.W. Pirkey Plant

Sample	Sample Date	Unit	Lithium Concentration
Bottom Ash (Solid Material)	2/11/2019	mg/kg	0.82 J
SPLP Leachate of Bottom Ash	2/11/2019	mg/L	0.011 J
EBAP Pond Water	6/2/2020	mg/L	0.0295
*WBAP Pond Water	11/4/2020	mg/L	0.0274
AD-31 - Average	May 2016 - November 2021	mg/L	0.0817
AD-32 - Average	May 2016 - November 2021	mg/L	0.1231

Notes:

mg/kg - milligram per kilogram

mg/L - milligram per liter

Average lithium values for monitoring wells AD-31 and AD-32 were calculated using all lithium data collected under 40 CFR 257 Subpart D, excluding statistically identified outliers.

J - Estimated value. Result is less than the reporting limit but greater than or equal to the method detection limit.

* - A sample was collected from the WBAP on 11/4/2020 as a surrogate for the EBAP, as the EBAP did not contain free water. The same process water is stored in both the WBAP and EBAP.

Table 5: Soil Lithium DataEast Bottom Ash Pond - H.W. Pirkey Plant

Location ID	Sample Depth (ft bgs)	Lithium (mg/kg)				
Bulk Soil Sample						
AD-32*	11	0.53				
AD-32*	20-25	1.60				
	10	5.30				
B-2	16	3.97				
В-2	71	7.42				
	87	13.10				
	10	3.64				
В-3	20	2.59				
	97	11.10				
Lignite	N/A	2.9 J				
Soli	d Material Retained After Filtr	ation				
AD-32*	13-33	9.8 J				
B-2	38-48	6.5 J				
B-3	29-34	7.8 J				
D-3	VAP 40-45	13.0				

Notes:

J - estimated value

mg/kg- milligram per kilogram

ft bgs - feet below ground surface

* - AD-32 samples were collected from a seperate borehole advanced near monitoring well AD-32

Depths for samples collected after filtration represent the screened interval for the permanent well where the sample was collected

VAP - vertical aquifer profiling

Table 6: Calculated Site-Specific Partition CoefficientsPirkey Plant - East Bottom Ash Pond

Source		B-2	Literature Value			
Unit	mg/L mg/kg L/kg			L/kg		
Element	Aqueous Phase	Adsorbed	Kd	Kd		
Li	0.081	6.5	80	43-370		
K	2.6	1100	423	42-1200		
Na	14	130	9	5.2-82		

Source		B-3	Literature Value	
Unit	mg/L	mg/kg L/kg		L/kg
Element	Aqueous Phase	Adsorbed	Kd	Kd
Li	0.097	7.8	80	43-370
K	2.9	1100	379	42-1200
Na	32	240	8	5.2-82

Source		AD-32*	Literature Value			
Unit	mg/L mg/kg L/kg			L/kg		
Element	Aqueous Phase			Kd		
Li	0.11	9.8	89	43-370		
K	3.9	1800	462	42-1200		
Na	57	220	4	5.2-82		

Notes:

mg/L: milligrams per liter

mg/kg: milligrams per kilogram

L/kg: liters per kilogram

Kd: partition coefficient

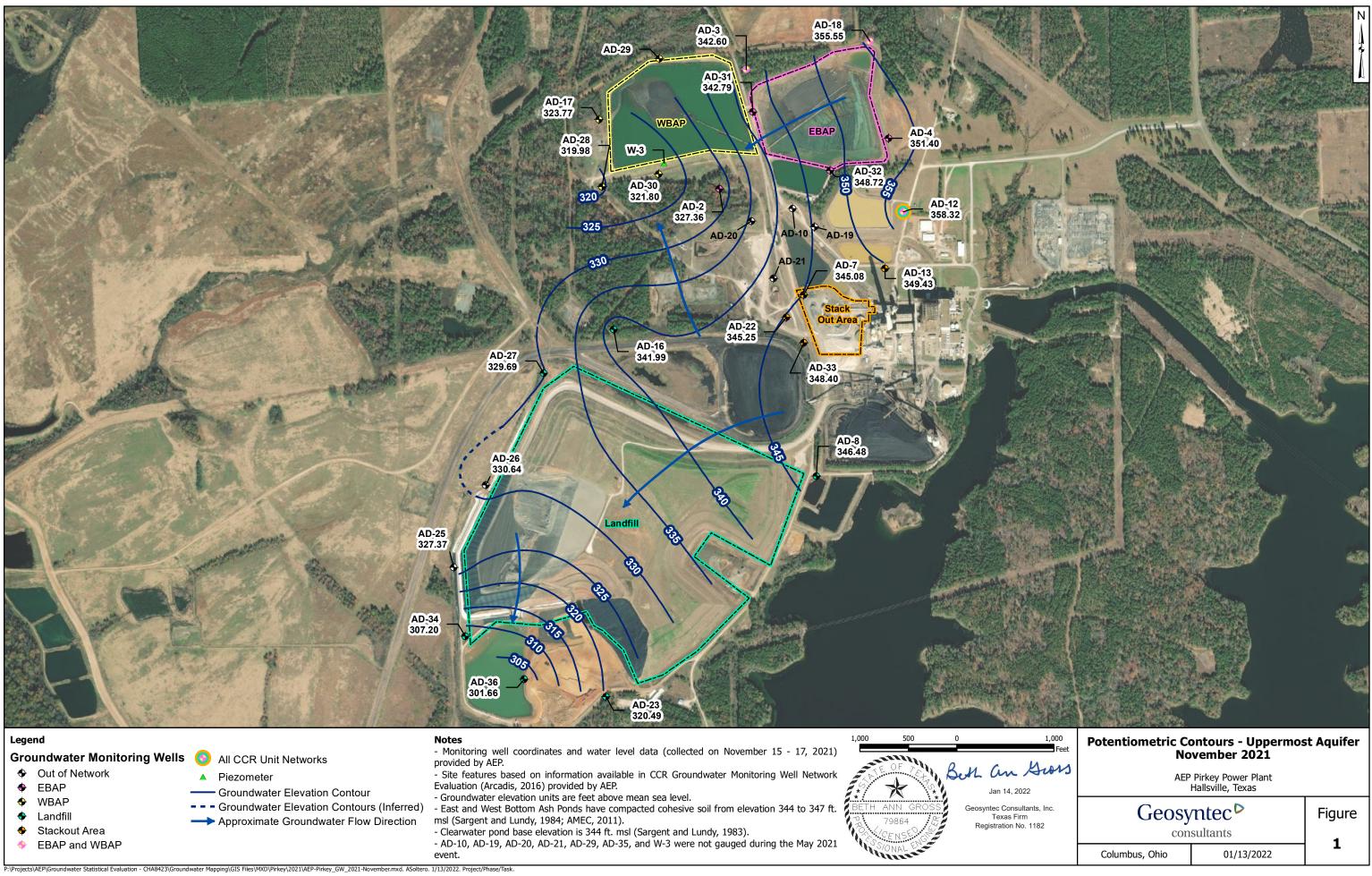
Adsorbed values are total metals concentrations reported by USEPA Method 6010B.

Literature values represent maximum and minimum values for the parameter as reported in Sheppard et al, 2009

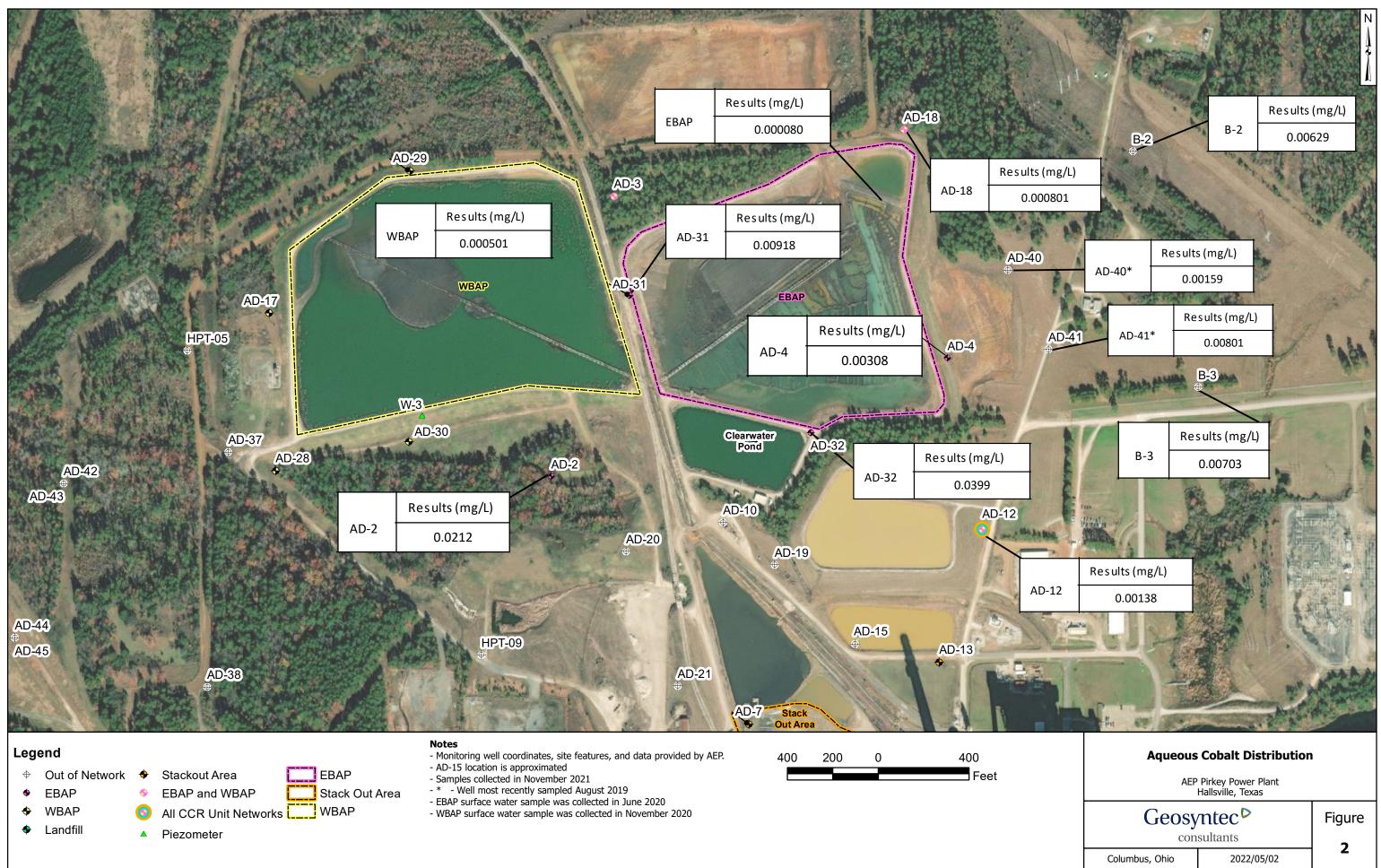
(Table 4-1, all sites) and Sheppard et al, 2011 (Table 3-3 cultivated peat and wetland peat only).

* - AD-32 samples were collected from a separate borehole advanced near monitoring well AD-32

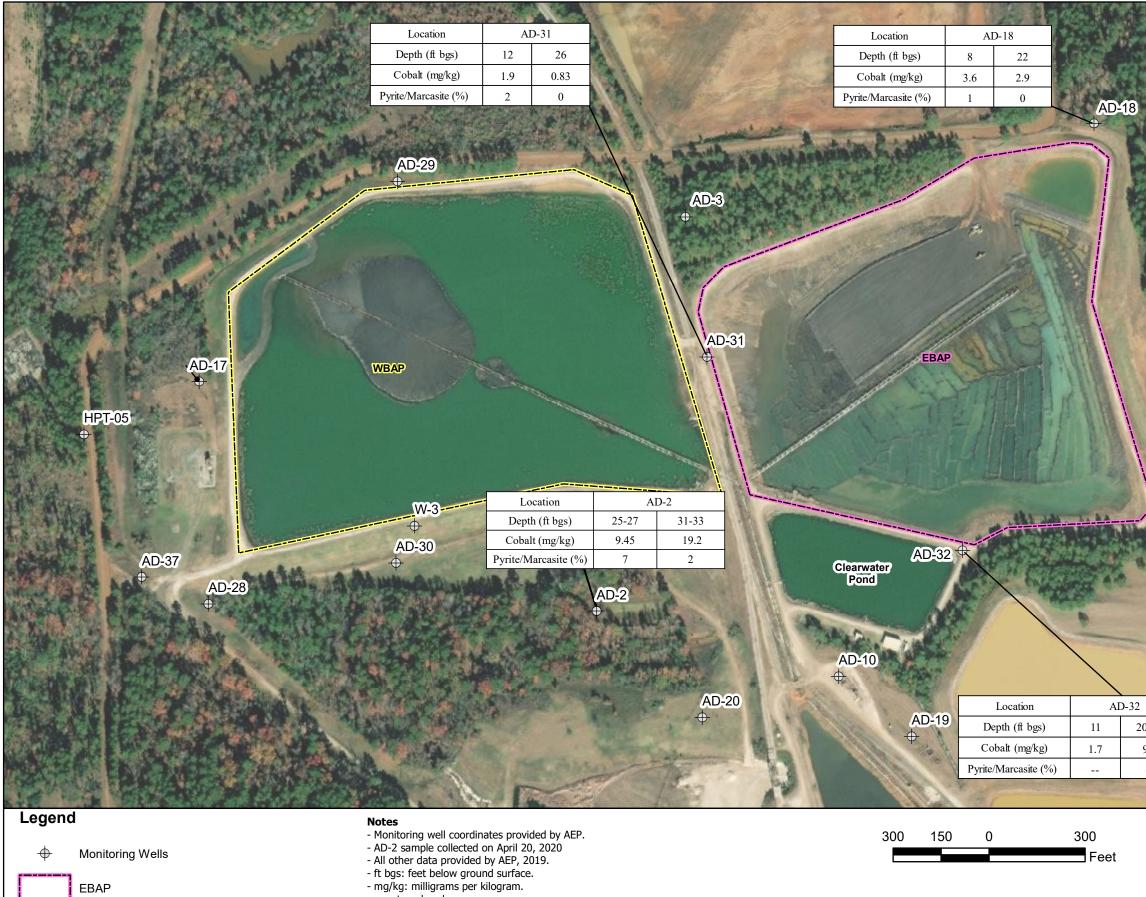
FIGURES







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EBAP

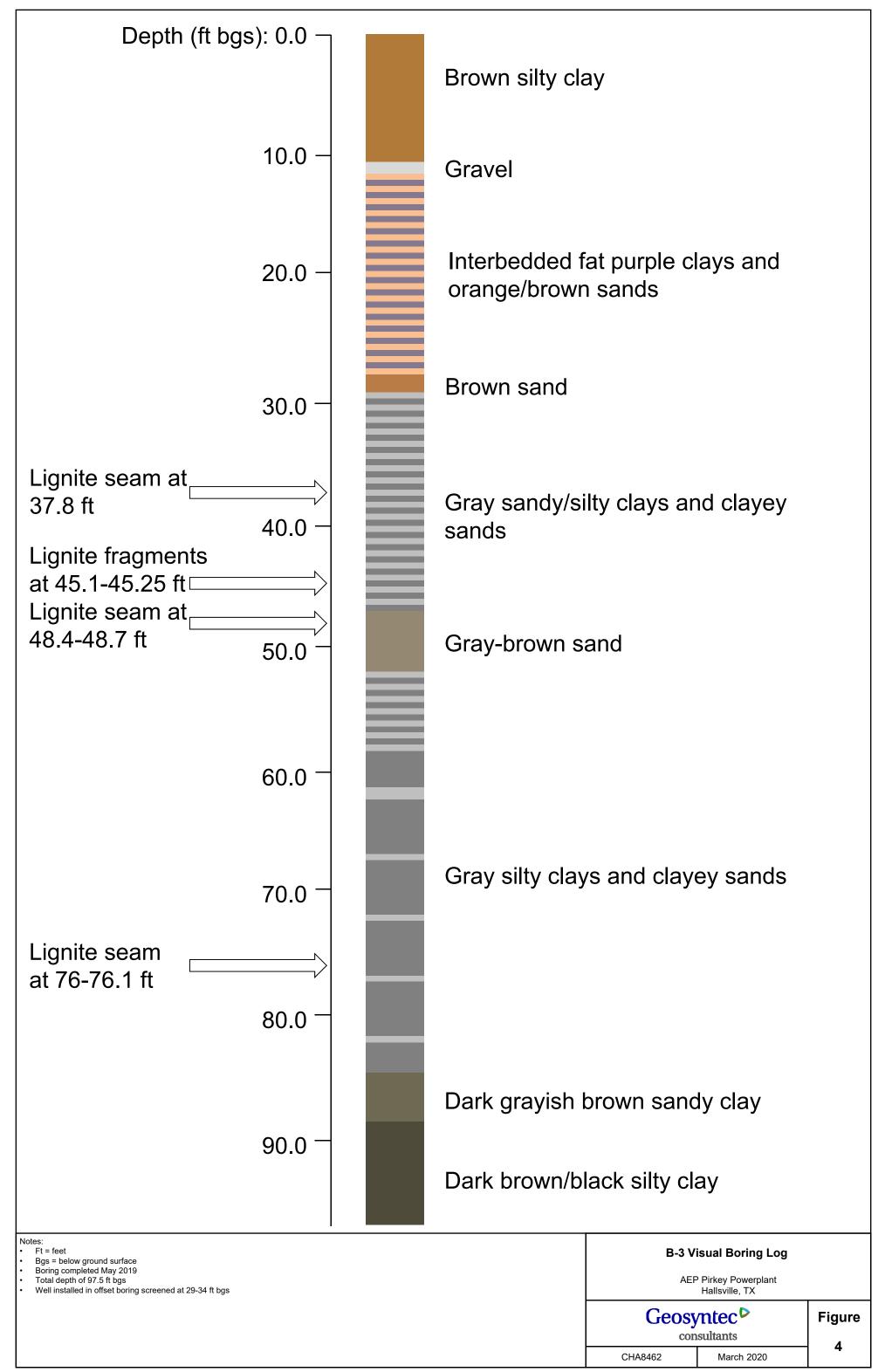
WBAP

- -- not analyzed.

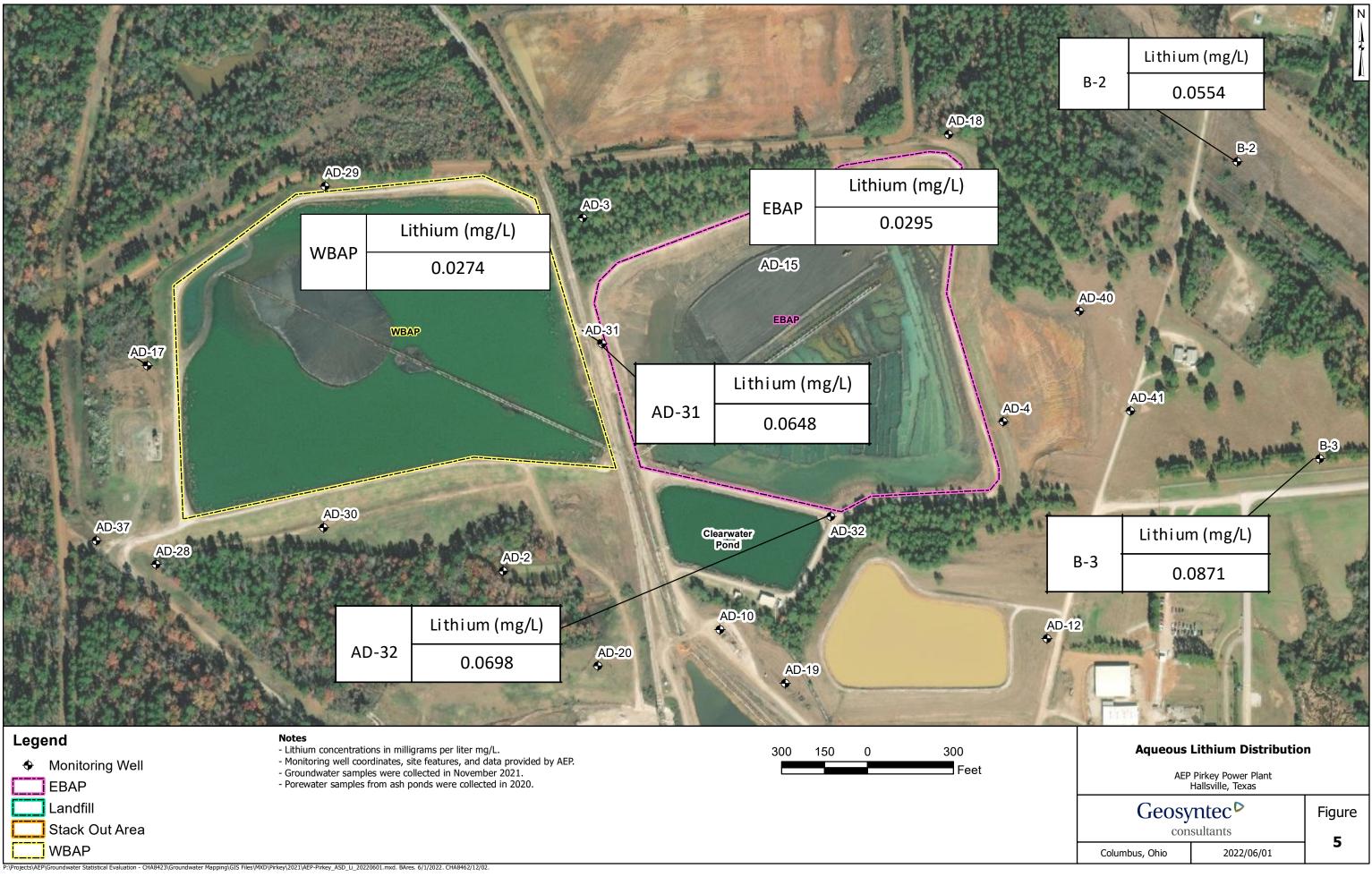
	1 1	1 1 A		1	COLONY !!	100	N
Location			B-2				
Depth (ft bgs)	10 16 71				82	87	₹
Cobalt (mg/kg)	2.36	3.62	10.30		7.21	3.11	L
Pyrite/Marcasite (%)	-	-		Contraction of the	-		
ALC: AN	1000						1
				F	3-2		.+1
	140	- An)- <u>-</u>		and the
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and the second		ocation			AD	9-41	
		th (ft bgs)		15	35		95
Marshall Stated	Cob	alt (mg/kg)	<	< 1.0 23.5		5	1.9
	Pyrite/N	Marcasite (%))	-	-		-
AD-4	ND-41 ∉	-	191	81x	hten	-	B-3
Sea as h >	Loc	ation		/	B-3	-	and the second s
taxa a a		(ft bgs)	10		20		97
and a start of		(mg/kg)	1.30		0.59		1.11
all and a		rcasite (%)	-		-		-
AD-12 () -25 -1 				- anti-		The ave	S. C. C.
	Cot	Dalt Dist AEP Pirke					

AEP Pirkey Power Plant Hallsville, Texas

Geosyntec▷ Figure consultants 3 Columbus, Ohio 2022/05/14

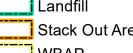


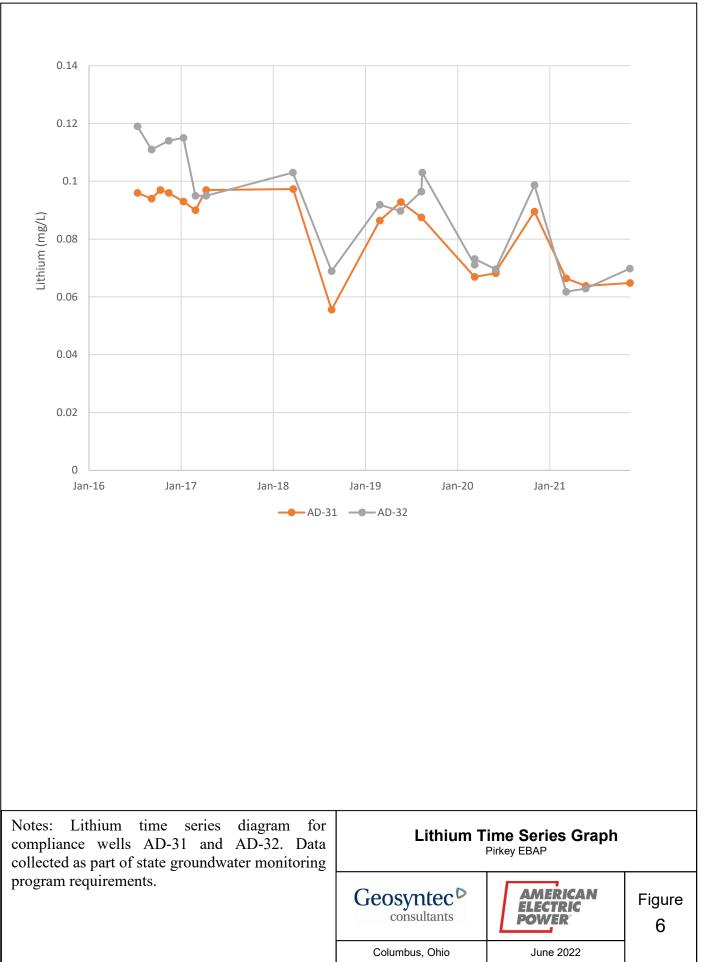
\\annarbor-01\data\Projects\AEP\Legal Department - ASD Review\Pirkey\2019-05 Field Investigation\Field Forms\Compiled Boring Logs\Visual boring logs





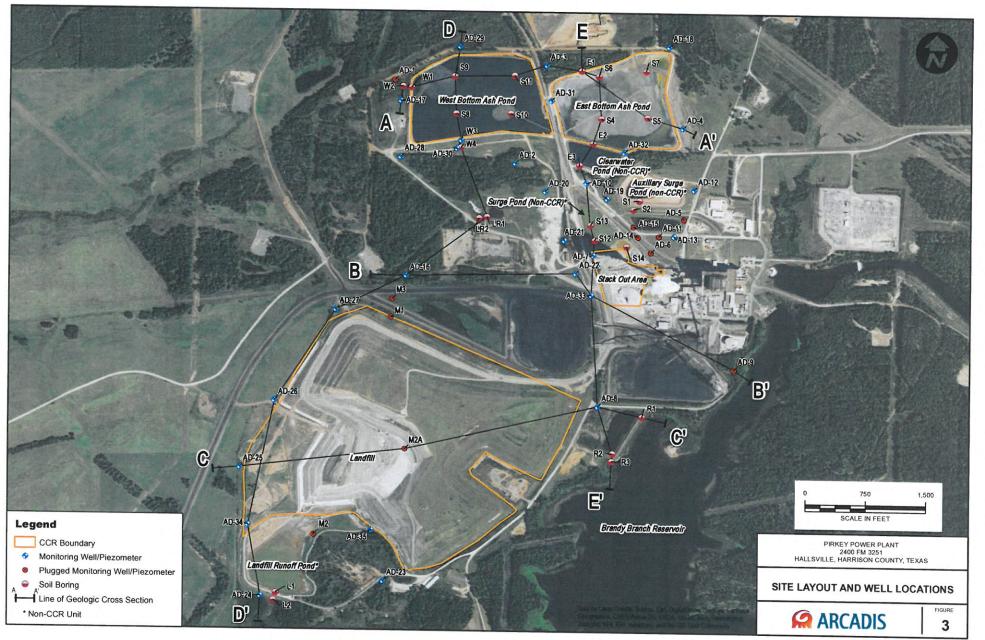


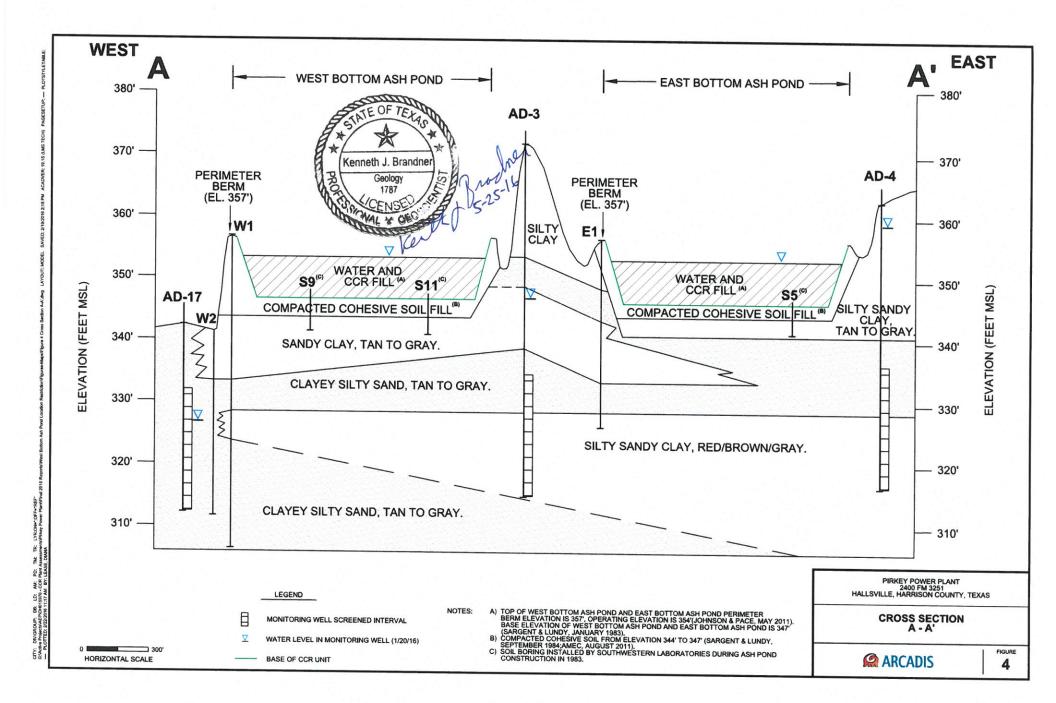




furk\2021-11 State ASD\Figures

ATTACHMENT A Geologic Cross-Section A-A'

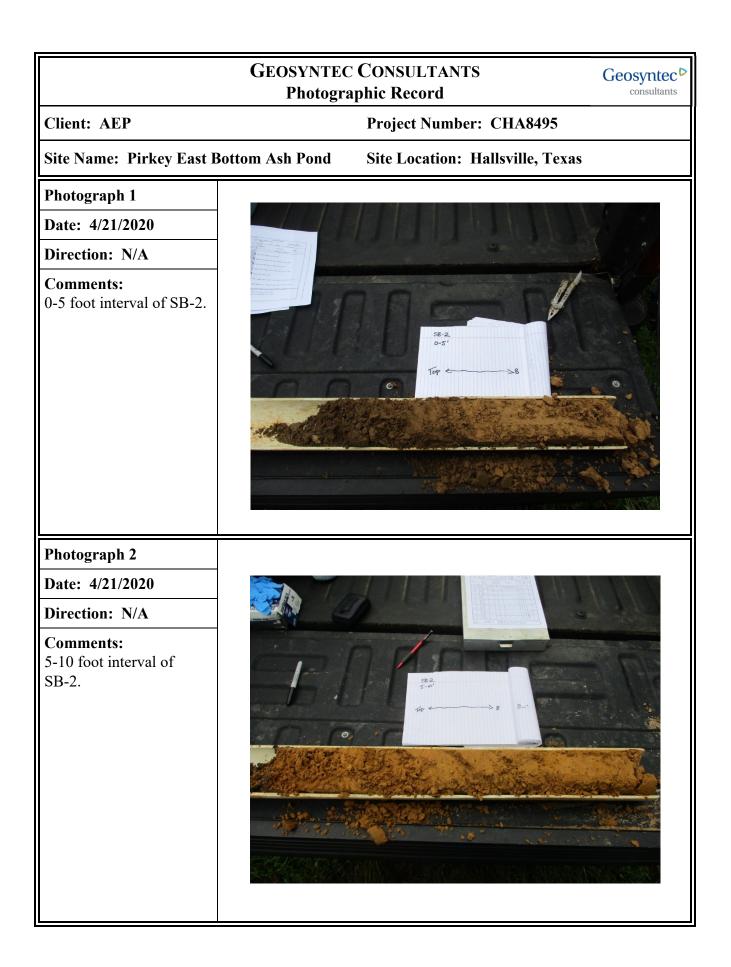




ATTACHMENT B SB-2 Boring Log

LOCATION A SILTS & SANDS CONDITION VLo Very Loose 0-4 Lo Loose 4-10 MDe Med. Dense 10-30 De Dense 30-50 VDe Very Dense >50			Vso So Mst St VSt	COH DNSISTENCY D. Very Soft Soft Soft Stiff Very Stiff Hard	0.5 - 1.0 4 1.0 - 2.0 8		4 G 8 T 15 R	COLORS LightBrBrown DarkBkBlack GreyBlBlue TanGrGrenn RedYYellow lish.Reddish.WhWhite	MATERIALS Cl Clay, Clayer Si Silt, Silty Sa Sand, Sandy Ls Limestone Gr Gravel SiS Siltstone SS Sandstone Sh Shale, Shaley	SAND A FFine MMedi CoCoar SiSilty		ine fedium oarse	Cale Calcare Lig Lignite Org Organi Lam Lamina	
Semple Interna TEST ASSIGNMENT	NO.	FT.	SAMPLES	~			1	CRIPTION	;	STANDA PENETRON			DIL	OR
	SAMPLE Recovery	DEPTH		CONDITION OR CONSISTENCY	COLOR	MINOR MATERIALS OR ADJECTIVES	PREDOMINATE MATERIAL	CHARACTE OR MODIFICA		SEAT - 6"	1st - 6"	2nd - 6"	UNIFIED SOIL CLASSIFICATION	N - VALUE OR HAND
0-5	2' Rec	0		0-81	Br. H. Rd Br	Si	Sa	Silty Sand-	souce clay,					
5-10	2.5'Rec				Lt. Rd.Br			- thin lenses (Jess them 1/4"				moist	(0-3
1D-15	4'RK	8.		8-142	H.Rd Ar. Fd Br, Gray	SALS'	. CI	Clay-some clayey say	dis intert	al de			moist	(10
.=	al R.	1115		, 05 Pc				in sand so	race iton one camise 10.5	912	112	.51	1. 10 B	
15-20	2'Rec	145		14.5-391	RLBY YI/W. Br. Gray	Si, C	52	STAY Sand-	and the second se	65			V-mais moist	the
20-25	* No Re		1		or in any		>(- centertet say	1 d c 20-25	3	ty)	VIMBIE	-(20
25-30	2.5 R				Bray-DKG DK.Br	mox ~	- AN	-gravel trement	d. partielly	and Cer	25 1evi	6	") set, "	-25'
					(25-5			- dark gray	1 sand e 25 tilty sat. saw	15	ally.	2.	- maist sati	25
30-35	3'Rec				Gra M		•	- sat silty sa	ud seame 3	21	1/1"		sat.	133. 32
35-40	4'REC	29		29-46	4.500.6	an CIS	</td <td>* some u.f. c between s chavey saidy</td> <td>typson ory</td> <td>rus</td> <td>(25</td> <td>-40</td> <td>exsand y v,n</td> <td>16154</td>	* some u.f. c between s chavey saidy	typson ory	rus	(25	-40	exsand y v,n	16154
				-91 19	4. 5ray,6 (39-46		- 01	-interbedde	t sitt i day	j@	39'-	846	m2175	- (39
					- N.			B.T.CHO'					<u></u>	
											1			
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								*31-33'	collected e in	35			A.	
Type Borii	ng HSA	Dr	y A	Auger ry Wasł		(OR) BA	AILED	25 ['] FT. WHILE D TOFT. UPO T AND CAVED 7	N COMPLET	ION		F	T. ON C	OMP

ATTACHMENT C SB-2 Boring Photographic Log



GEOSYNTEC CONSULTANTS Photographic Record

Geosyntec Consultants

Client: AEP

Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond

Site Location: Hallsville, Texas

Photograph 3

Date: 4/21/2020

Direction: N/A

Comments:

10-15 foot interval of SB-2.



Photograph 4

Date: 4/21/2020

Direction: N/A

Comments:

15-20 foot interval of SB-2. Recovery of this interval was limited.



GEOSYNTEC CONSULTANTS Photographic Record



Client: AEP

Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond

Site Location: Hallsville, Texas

Photograph 5

Date: 4/21/2020

Direction: N/A

Comments:

20-25 foot interval of SB-2. Recovery of this interval was limited.



Photograph 6

Date: 4/21/2020

Direction: N/A

Comments:

25-30 foot interval of SB-2. Very little of this interval was recovered. A color change was observed from red to dark brown/black. A sample was collected from this interval.



GEOSYNTEC CONSULTANTS Photographic Record

Geosyntec^D consultants

Client: AEP

Project Number: CHA8495

Site Name: Pirkey East Bottom Ash Pond

Site Location: Hallsville, Texas

Photograph 9

Date: 4/21/2020

Direction: N/A

Comments:

30-35 foot interval of SB-2. Very little of this interval was recovered.. A sample was collected from this interval.



Photograph 10

Date: 4/21/2020

Direction: N/A

Comments: 35-40 foot interval of SB-2



ATTACHMENT D SEM/EDS Analysis



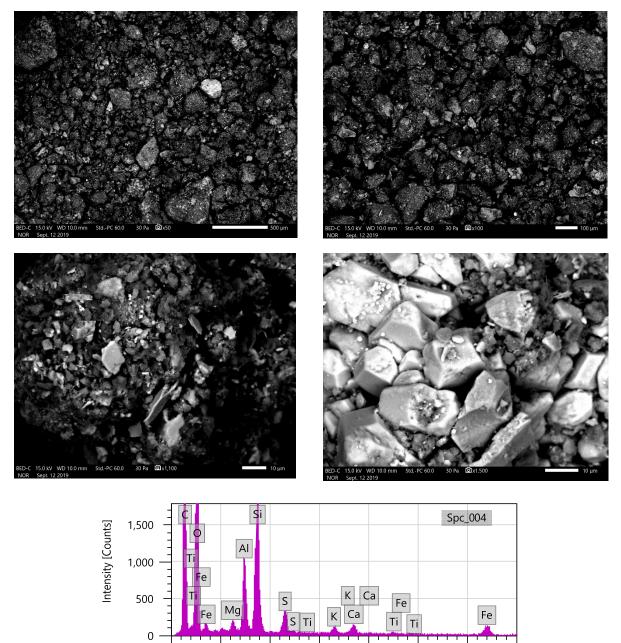
September 16, 2019

Dr. Bruce Sass

941 Chatham Lane, Suite 103, Columbus, OH 43221

0

via Email: <u>BSass@geosyntec.com</u> bus, OH 43221



Lignite. Backscattered electron micrographs show the sample at 100X, 1,100X, and 1,500X. EDS spectrum at bottom is an area scan of the region shown in top right micrograph. Bright particles are mostly quartz and feldspar. Major peaks for carbon, oxygen, silicon, and aluminum suggest coal and clay.

3

Energy [keV]

5

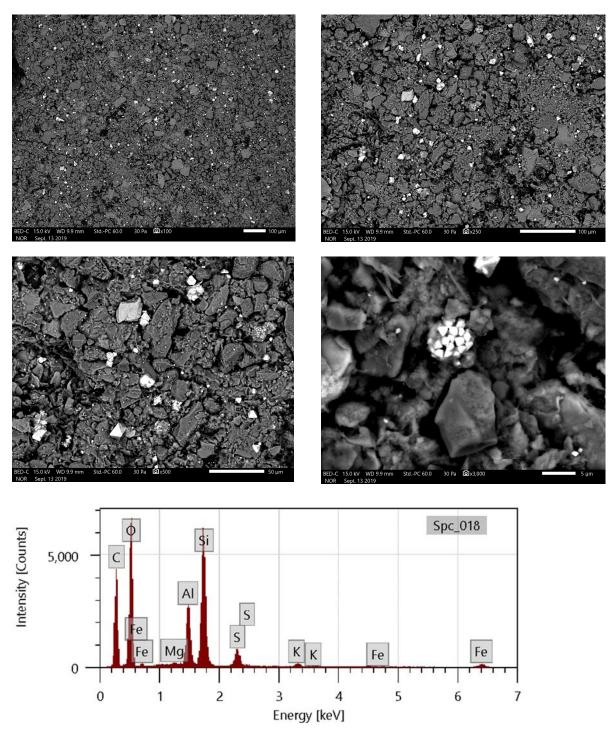
6

7

2

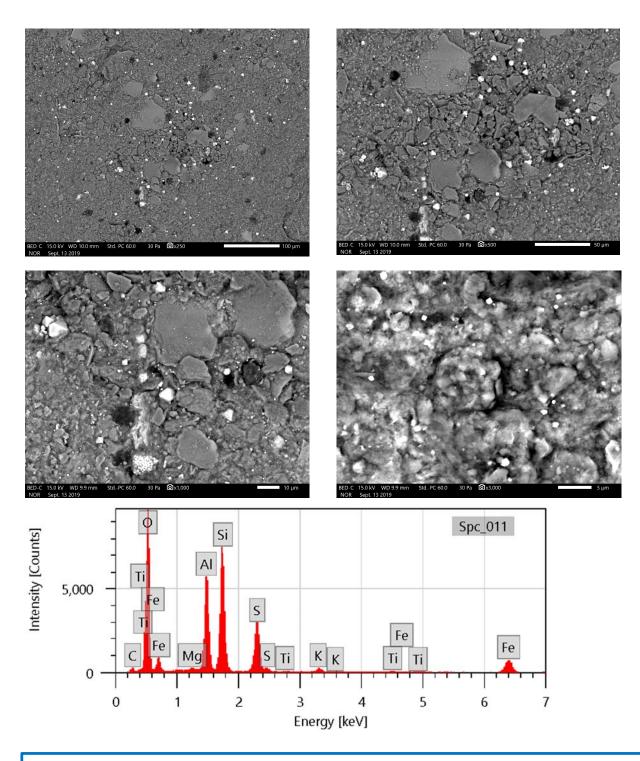
Austin, TX • Chicago, IL • Washington, DC • Doha, Qatar

Corporate Office: 5400 Old Orchard Road, Skokie, IL 60077-1030 P: 847-965-7500 F: 847-965-6541 www.CTLGroup.com CTLGroup is a registered d/b/a of Construction Technology Laboratories, Inc.



Sample VAP B3 40-45. Backscattered electron micrographs show the sample at 100X, 250X, 500X, and 3000X. EDS spectrum at bottom is an area scan of the region shown at 500X. Bright particles are pyrite (framboid in bottom right micrograph). Major peaks for carbon, oxygen, silicon, and aluminum suggest coal and clay.





Sample VAP B3 50-55. Backscattered electron micrographs show the sample at 250X, 500X, 1000X, and 3000X. EDS spectrum at bottom is an area scan of the region shown at 3000X. Bright particles are mostly pyrite (framboid in bottom left micrograph); occasional particles of Fe-Ti oxide are detected. Major peaks for oxygen, silicon, and aluminum suggest clay. Large blocky particles are mostly quartz, feldspar, and clay.



ATTACHMENT E Certification by a Qualified Professional Engineer

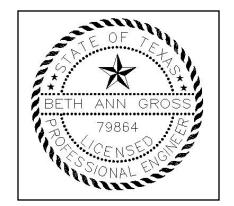
CERTIFICATION BY A QUALIFIED PROFESSIONAL ENGINEER

I certify that the selected and above described alternative source demonstration is appropriate for evaluating the groundwater monitoring data for the Pirkey East Bottom Ash Pond CCR management area and that the requirements of 30 TAC § 352.951(e) have been met.

Beth Ann Gross Printed Name of Licensed Professional Engineer

Signature

79864 License Number Texas Licensing State <u>June 16,2022</u> Date



Geosyntec Consultants 2039 Centre Pointe Blvd, Suite 103 Tallahassee, Florida 32308 Texas Registered Engineering Firm No. F-1182